# Xestoproxamines A-C from Neopetrosia proxima. Assignment of Absolute Stereostructure of Bis-piperidine Alkaloids by Integrated Degradation-CD Analysis 

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(S) Supporting Information


#### Abstract

The complete stereostructures of xestoproxamines A-C, from the Bahamian sponge Neopetrosia proxima, were assigned from spectroscopic analysis, including MS, 2D NMR, and integrated degradation-CD analysis. Two new $C D$ application protocols are described for defining absolute configuration: one for allylic methyl groups in branched chains and a second for the heterocyclic core bis-piperidine with specific applicability to other members of this class alkaloids-known for their stereohetero-geneity-and tertiary cyclic amines in general.




Polycyclic amines comprise a family of biologically active macrocycles reported from sponges of the genera Haliclona, Xestospongia, Petrosia, Neopetrosia, and Reniera, among others. ${ }^{1}$ Their structures contain two or more nitrogen atoms woven into networks of fused heterocyclic and carbocyclic rings, mainly based on piperidine or cyclohexane. From the standpoint of structural characterization, polycyclic amines present difficulties in their purification due to their amphiphilic physical properties, while structure elucidation is hampered by both limited proton chemical shift dispersion presented in their ${ }^{1} \mathrm{H}$ NMR spectra and stereochemical heterogeneity. In fact, the structures of several alkaloids in this class have succumbed only to X-ray crystallography. We report here three new bis-piperidine alkaloids, xestoproxamines A-C $(1-3)$, from Neopetrosia proxima ${ }^{2}$ and assignment of their complete absolute stereostructures by an integrated approach based on NMR and circular dichroism (CD). ${ }^{3}$ A notable advancement is chiroptical analysis by CD, following double quaternization of the bis-piperidine, to give $N$, $N^{\prime}$-bis-bromophenacyl derivatives that display characteristic split Cotton effects that reliably reflect the absolute configuration of the bis-piperidine core.

The molecular formula for 1 was established as $\mathrm{C}_{30} \mathrm{H}_{52} \mathrm{~N}_{2}$ based on HRESIMS data $\left(m / z\right.$ 441.4207, $\left.[\mathrm{M}+\mathrm{H}]^{+}\right)$. The presence of two 1,2-disubstituted carbon - carbon double bonds was supported by two coupled pairs of $\mathrm{CH}=\mathrm{CH}$ spin systems in the $1 \mathrm{H}-1 \mathrm{H}$ COSY spectrum (Table $1, \delta 5.49 \mathrm{~m} ; 5.37 \mathrm{~m} ; 5.31, \mathrm{~m}$, and $5.28, \mathrm{~m}$ ) and four $\mathrm{CH} \mathrm{sp}{ }^{2}$ carbons in the ${ }^{13} \mathrm{C}$ DEPT spectrum ( $\delta 131.74, \mathrm{~d} ; 131.69, \mathrm{~d} ; 130.1, \mathrm{~d}$, and 129.1, d). NOESY correlations observed between allylic methylene groups ( $\mathrm{H} 16_{\mathrm{b}} / \mathrm{H} 19_{\mathrm{b}}$ and $\mathrm{H} 26_{\mathrm{b}} / \mathrm{H} 29_{\mathrm{b}}$ ) confirmed the $Z$-configuration of both double
bonds. Because all $\mathrm{sp}^{2}$ carbons were accounted for and no $\mathrm{C}=\mathrm{N}$ bonds were present, the remaining N atoms must be $\mathrm{sp}^{3}$ hybridized; therefore, $\mathbf{1}$ is a ditertiary amine related to the known alkaloids halicyclamines A $(4)^{4}$ and B (5), ${ }^{5}$ haliclonacyclamines A (6) and B (7), ${ }^{6}$ arenosclerin A (8), and haliclonacyclamine E (9). ${ }^{7}$ The remaining four degrees of unsaturation were completed by four rings. Interpretation of ${ }^{13} \mathrm{C}, ~ D E P T$, and HSQC data showed six N -substituted $\mathrm{CH}_{2}$ groups ( $\delta 58.5,56.9,56.4$, $55.3,48.9$, and 47.4) that were linked to the remaining ring and chain elements (substructures a-c, Figure 1), including two substituted piperidine rings, from additional 2D NMR evidence (DQF-COSY, TOCSY, gHSQC, and gHMBC), in particular, HMBC cross-peaks of H3/C9, H4/C9, and H10/C3. Contiguous spin systems identified by TOCSY cross-peaks of H13b/ H 16 b and $\mathrm{H} 11 \mathrm{~b} / \mathrm{H} 14 \mathrm{~b}$ secured two long hydrocarbon linking chains between the $\mathrm{CH}_{2}-\mathrm{N}$ and $\mathrm{CH}-\mathrm{N}$ groups. Substructure c was joined to a by HMBC cross-peaks from H 2 to both C 29 and C30. The final ring was closed by connecting substructure $\mathbf{b}$ to $\mathbf{c}$ through HMBC correlations from H 22 to C24.

The relative configuration of the conjoined bis-piperidine rings in $\mathbf{1}$ (designated rings A and B, Figure 2) could be established from $J$ coupling constants and NOESY data. The relative orientation of $\mathrm{H} 2_{\mathrm{ax}}$ and $\mathrm{H} 3_{\mathrm{ax}}$ was established by a 1D-TOCSY experiment; irradiation of $\mathrm{H5}_{\mathrm{eq}}(\delta 3.48, \mathrm{brd})$ showed that $\mathrm{H1}_{\mathrm{ax}}(\mathrm{dd}, J=12.0$, $12.0 \mathrm{~Hz})$ revealed two large couplings to $\mathrm{H1}_{\mathrm{eq}}(\delta 3.18, \mathrm{~m})$ and

[^0]
xestoproxamine $A(1)-\Delta^{27}$
xestoproxamine $B$ (2)

halicyclamine A (4)

(-)-haliclonacyclamine $A(6): Z-\Delta^{25}$ (+)-haliclonacyclamine B(7): Z- $\Delta^{29}$

(-)-perhydrohaliclonacyclamine (26) $13 E, 15 Z, 23 Z, 25 Z^{-2} 13,15,23,25(+)-(31)$
arenosclerin $A: Z-, \mathrm{R}=\mathrm{OH}(8)$
haliclonacyclamine $\mathrm{E}: \mathrm{Z}, \mathrm{R}=\mathrm{R}=\mathrm{H})$ haliclonacyclamine $E: Z-, R=H(9)$
perhydrohaliclonacyclamine $E:(28)$

xestoproxamine $C$ (3)

halicyclamine $B(5)$


(-)-neopetrosiamine (32)



Figure 1. Substructures $\mathbf{a}-\mathbf{c}$ of xestoproxamine $A$ (1). Points of attachment of linking chains are indicated by $\bullet$.
to $\mathrm{H} 2_{\mathrm{ax}}(\delta 2.29, \mathrm{~m})$ respectively. In addition, $\mathrm{H} 3_{\mathrm{ax}}$ was vicinally coupled by two large scalar couplings $(J=11.0 \mathrm{~Hz})$ to $\mathrm{H} 2_{\mathrm{ax}}$ and $\mathrm{H} 4_{\mathrm{ax}}$. Additional large couplings $(J=12.0$ and 12.0 Hz , respectively) from $\mathrm{H} 7_{\mathrm{ax}}$ and $\mathrm{H} 9_{\mathrm{ax}}$ to $\mathrm{H} 8_{\mathrm{ax}}$ ( $\delta 1.06$, ddd, $J=$ $12.0,12.0,12.0 \mathrm{~Hz})$ showed the three protons to be axially disposed. The relative configuration of the remaining stereocenters in 1 was assigned by interpretation of NOESY correlations. The dihedral angle of $\sim 90^{\circ}$ between H3 and H9 was supported by lack of vicinal coupling ( ${ }^{3} J \approx 0 \mathrm{~Hz}$ ). Conformational constraints placed on the rings by NOE and $J$ data, in particular, mutual correlations between the pairs $\mathrm{H} 2_{\mathrm{ax}}$ and $\mathrm{H} 8{ }_{\mathrm{ax}}$
and $\mathrm{H}_{10} \mathrm{eq}_{\mathrm{eq}}$ and $\mathrm{H} 3_{\mathrm{ax}}$, led to the depicted relative configuration and a conformer that places the averaged planes of ring A and B orthogonal to each other (Figure 2).

Xestoproxamine $B$ (2) showed a molecular formula of $\mathrm{C}_{30} \mathrm{H}_{54} \mathrm{~N}_{2}$, which has one degree of unsaturation less than xestoproxamine A (1). The ${ }^{1} \mathrm{H}$ NMR signals for 2 (Table 2) were similar to those of 1 except for the presence of two vinyl protons ( $\delta 5.48$, m and $5.33, \mathrm{~m}$ ), indicating only one $\mathrm{C}=\mathrm{C}$ double bond. 2D NMR data confirmed both $\mathbf{1}$ and 2 share the same bispiperidine ring system, including relative configuration ( $J_{\mathrm{HH}}$ and NOESY). The double bond was positioned between C17/C18 by COSY and TOCSY correlations to H19. Therefore xestoproxamine $B(2)$ is a dihydro derivative of xestoproxamine A (1).

The third new bis-piperidine alkaloid, xestoproxamine C (3), $\mathrm{C}_{31} \mathrm{H}_{56} \mathrm{~N}_{2}$, showed a characteristic signal for a secondary methyl branch in the ${ }^{1} \mathrm{H}$ NMR spectrum (Table 3, $\delta 0.87, \mathrm{~d}, J=6.7 \mathrm{~Hz}$ ) that was absent in 1 and 2.2D NMR and NOESY data indicated the configuration of 3 in the core bis-piperidine heterocycle differed from $\mathbf{1}$ and $\mathbf{2}$ but corresponded to the relative configuration found in haliclonacyclamine A (6). ${ }^{5}$ While ring B remained in the same conformation as $\mathbf{1}$ and 2 , ring $A$ in 3 had undergone ring-flip to the opposite chair conformation, apparently without inversion at the ring $\mathrm{A} \mathrm{sp}^{3}$ nitrogen (NOESY, Figure 2), a consequence of the inverted C2 configuration; yet in the new torsional arrangement a dihedral angle of $\sim 90^{\circ}$ for H3$\mathrm{C} 3-\mathrm{C} 9-\mathrm{H} 9\left({ }^{3} \mathrm{~J}_{\mathrm{H} 3-\mathrm{H} 9} \approx 0 \mathrm{~Hz}\right)$ was retained.

Compound 3 failed to produce suitable X-ray quality crystals under a variety of conditions. Stereoanalysis by spectroscopic means would require separate assignments of the two stereoelements in 3: the C23 stereocenter and the bis-piperidine core. We devised the following degradative approach for assignment of the lone C23 stereocenter exo to the heterocyclic core. Hofmann elimination of suitably quaternized nitrogen atoms in 3 would provide a terminal olefin ( $\Delta^{21}$ ), which could then be subjected to cross metathesis with a styrenyl derivative to insert a chromophore next to the secondary Me branch (first sphere of asymmetry), rendering the molecule amenable to analysis by CD. Due to the scarcity of 3 , the method was first piloted with more abundant xestoproxamine A (1) (Scheme 1). Catalytic hydrogenation of $\mathbf{1}\left(\mathrm{Pd}-\mathrm{C}, \mathrm{MeOH}, \mathrm{H}_{2}\right)$, purification and conversion to the free base $\mathbf{1 0}$ by HPLC with buffered solvent (Lux-cellulose; $\mathrm{CH}_{3} \mathrm{CN} / i$ - $\mathrm{PrOH} / \mathrm{Et}_{2} \mathrm{NH}$ ), and immediate exhaustive alkylation with $\mathrm{CH}_{3} \mathrm{I}$ cleanly provided the bis-methiodide salt 11 . The iodide counterion was exchanged with hydroxide by treatment of $\mathbf{1 1}$ with either $\mathrm{Ag}_{2} \mathrm{O}$ or strong anion exchange resin (Amberlite-IRA 400, $\mathrm{HO}^{-}$form). Two possible major products were anticipated from cleavage of $\mathrm{C} 21-\mathrm{N}$ and cleavage of either $\mathrm{C} 5-\mathrm{N}$ or $\mathrm{C} 11-\mathrm{N}$, based on the Hofmann rule that leads to the least substituted alkenes. In the event, Hofmann elimination ( $10 \mathrm{~min}, 300 \mathrm{~W}, 140^{\circ} \mathrm{C}$ ) of the neat quaternary ammonium hydroxide double salt afforded only one major alkene product, 12. The structure of $\mathbf{1 2}$ was confirmed by 2D NMR (COSY, TOCSY, HSQC, and HMBC), which showed piperidine ring B was intact: exo cleavage of the $\mathrm{C} 5-\mathrm{N}$ bond was supported by the multiplicity of $\mathrm{H} 4(\delta 5.55, \mathrm{dt}, J=17.0,10.3 \mathrm{~Hz})$. In contrast, ring A had undergone endo cleavage of the $\mathrm{C} 21-\mathrm{N}$ bond, and the vinyl methine signal exhibited the more complex pattern for vinyl proton H 22 associated with a terminal allyl group ( $\delta 5.80$, dddd, $J=17.0,10.3,6.8,6.8 \mathrm{~Hz}, 1 \mathrm{H}$ ). Alkene 12 was subjected to cross metathesis with excess 2 -methoxy-6vinylnaphthalene $13^{8}$ ( $\sim 10$ equiv) in the presence of Grubbs' second-generation catalyst, ${ }^{9}$ which selectively engaged only the

Table 1. ${ }^{1} \mathrm{H}(600 \mathrm{MHz})$ and ${ }^{13} \mathrm{C}$ NMR ( 125 MHz ) for 1 $\left(\mathrm{CD}_{3} \mathrm{OD}\right)$

| no. | $\delta_{\mathrm{C}}$, mult. ${ }^{\text {a }}$ | $\begin{gathered} \delta_{\mathrm{H}}, \text { mult } \\ (J \text { in } \mathrm{Hz}, \mathrm{ax} / \mathrm{eq}) \end{gathered}$ | $\begin{aligned} & \text { DQF- } \\ & \text { COSY } \end{aligned}$ | $\mathrm{HMBC}^{\text {b }}$ |
| :---: | :---: | :---: | :---: | :---: |
| 1 | 55.3, $\mathrm{CH}_{2}$ | $\begin{aligned} & 3.32, \mathrm{dd}(12.0, \mathrm{ax})^{e} \\ & 3.18, \mathrm{~m}(\mathrm{eq}) \end{aligned}$ | 2 | $\begin{aligned} & 2,3,11,30 \\ & 2,3,5,11,30 \end{aligned}$ |
| 2 | 29.7, CH | 2.29, m (ax) | 1, 3, 30 | 3, 29, 30 |
| 3 | 36.0, CH | 1.91, m (ax) | 2, 4, 4 | $4,5,8,9,10,30$ |
| 4 | 25.2, $\mathrm{CH}_{2}$ | $\begin{aligned} & 1.80, \mathrm{~m}(\mathrm{eq}) \\ & 1.69, \mathrm{~m}(\mathrm{ax}) \end{aligned}$ | 3 | $\begin{aligned} & 3,5,9, \\ & 2,3 \end{aligned}$ |
| 5 | 47.4, $\mathrm{CH}_{2}$ | $\begin{aligned} & 3.48, \operatorname{brd}(13.2, \mathrm{eq}) \\ & 3.19, \mathrm{~m}(\mathrm{ax}) \end{aligned}$ | 4 | $\begin{aligned} & 1,3,4,11 \\ & 1,3,4 \end{aligned}$ |
| 6 | 58.5, $\mathrm{CH}_{2}$ | $3.21, \operatorname{brd}(12.0, \mathrm{eq})$ <br> 2.84 , dd (12.0, 12.0, 12.0, ax) | 7 | $\begin{aligned} & 7,8,9,10,20,21 \\ & 7,8,9,10 \end{aligned}$ |
| 7 | 34.3, CH | 1.90, m (ax) | 6, 8, 20 | 6, 8, 9, 10, 20 |
| 8 | 27.5, $\mathrm{CH}_{2}$ | $\begin{aligned} & 1.84, \mathrm{~m}(\mathrm{eq}) \\ & 1.06 \text {, } \mathrm{ddd}(12.0,12.0,12.0 \mathrm{ax}) \end{aligned}$ | 7,9 | $\begin{aligned} & 3,6,7 \\ & 6,7,20 \end{aligned}$ |
| 9 | 42.1, CH | 1.74, m (ax) | 8, 10 | 2, 3, 4, 7, 8 |
| 10 | 56.9, $\mathrm{CH}_{2}$ | $\begin{aligned} & 3.69, \mathrm{~m}(\mathrm{eq}) \\ & 2.84, \mathrm{dd}(12.0,12.0 \mathrm{ax}) \end{aligned}$ | 9 | $\begin{aligned} & 3,6,8,9,21 \\ & 6,8,9 \end{aligned}$ |
| 11 | 48.9 ${ }^{\text {c }} \mathrm{CH}_{2}$ | $\begin{aligned} & 3.32, \mathrm{~m} \\ & 3.02, \mathrm{ddd}(12.8,12.8,3.7) \end{aligned}$ | 12 | $\begin{aligned} & 1,5,12,13 \\ & 5,12,13 \end{aligned}$ |
| 12 | 25.2, $\mathrm{CH}_{2}$ | $\begin{aligned} & 1.79, \mathrm{~m} \\ & 1.66, \mathrm{~m} \end{aligned}$ | 11, 13 | $\begin{aligned} & 11,13,14 \\ & 11,14 \end{aligned}$ |
| 13 | 21.7, $\mathrm{CH}_{2}$ | $\begin{aligned} & 1.79, \mathrm{~m} \\ & 1.65, \mathrm{~m} \end{aligned}$ | 12, 14 | $\begin{aligned} & 11,12,14,15 \\ & 11,12,14 \end{aligned}$ |
| 14 | 27.3, $\mathrm{CH}_{2}$ | $\begin{aligned} & 1.52, \mathrm{~m} \\ & 1.44, \mathrm{~m} \end{aligned}$ | 13, 15 | $\begin{aligned} & 12,13,15,16 \\ & 12,13,15,16 \end{aligned}$ |
| 15 | 29.0, $\mathrm{CH}_{2}$ | $\begin{aligned} & 1.43, \mathrm{~m} \\ & 1.25, \mathrm{~m} \end{aligned}$ | 14, 16 | $\begin{aligned} & 14,16,17 \\ & 14,16,17 \end{aligned}$ |
| 16 | 29.4, $\mathrm{CH}_{2}$ | $\begin{aligned} & 2.25, \mathrm{~m} \\ & 2.00, \mathrm{~m} \end{aligned}$ | 15, 17 | $\begin{aligned} & 15,17,18 \\ & 14,17,18 \end{aligned}$ |
| 17 | $131.69^{d}$, CH | 5.49, m | 16, 18 | 15, 16, 18, 19 |
| 18 | 129.1, CH | 5.31, m | 17, 19 | 16, 19, 20, |
| 19 | 23.4, $\mathrm{CH}_{2}$ | $\begin{aligned} & \text { 2.50, m } \\ & \text { 2.12, brd (15.4) } \end{aligned}$ | 18, 20 | $7,17,18,20$ |
| 20 | 31.5, $\mathrm{CH}_{2}$ | $\begin{aligned} & 1.74, \mathrm{~m} \\ & 1.27, \mathrm{~m} \end{aligned}$ | 7 | $\begin{aligned} & 6,7,8,18,19 \\ & 6,7,8 \end{aligned}$ |
| 21 | 56.4, $\mathrm{CH}_{2}$ | 3.20, m | 22 | 6, 22, 23 |
| 22 | 22.6, $\mathrm{CH}_{2}$ | 1.81, m | 21, 23 | 21,23, 24, 25 |
| 23 | 28.1, $\mathrm{CH}_{2}$ | $\begin{aligned} & 1.62, \mathrm{~m} \\ & 1.42, \mathrm{~m} \end{aligned}$ | 22, 24 | $\begin{aligned} & 21,22,24,25 \\ & 21,22,24 \end{aligned}$ |
| 24 | 27.0, $\mathrm{CH}_{2}$ | 1.42, m | 23, 25 | 22, 23, 25, 26 |
| 25 | 29.0, $\mathrm{CH}_{2}$ | $\begin{aligned} & 1.53, \mathrm{~m} \\ & 1.45, \mathrm{~m} \end{aligned}$ | 24, 26 | $\begin{aligned} & 23,24,26,27 \\ & 23,24,26,27 \end{aligned}$ |
| 26 | 27.1, $\mathrm{CH}_{2}$ | $\begin{aligned} & 2.35, \mathrm{~m} \\ & 1.91, \mathrm{~m} \end{aligned}$ | 25, 27 | $\begin{aligned} & 28 \\ & 24,25,27,28 \end{aligned}$ |
| 27 | $131.74{ }^{\text {d }}$, CH | 5.37, m | 26, 28 | 25, 26, 28, 29 |
| 28 | 130.1, CH | 5.28, m | 27, 29 | 26, 27, 29, 30 |
| 29 | 21.5, $\mathrm{CH}_{2}$ | $\begin{aligned} & 2.24, \mathrm{~m} \\ & 1.82, \mathrm{~m} \end{aligned}$ | 28, 30 | $\begin{aligned} & 28,30 \\ & 27,28,30 \end{aligned}$ |
| 30 | 31.9, $\mathrm{CH}_{2}$ | $\begin{aligned} & 1.72, \mathrm{~m} \\ & 1.57, \mathrm{~m} \end{aligned}$ | 29 | $\begin{aligned} & 1,3,28,29 \\ & 1,3,28,29 \end{aligned}$ |

${ }^{a}$ Determined from DEPT and HSQC. ${ }^{b}$ HMBC correlations, optimized for 8 Hz , are from proton(s) stated to the indicated carbon. ${ }^{c}$ Observed by HSQC. ${ }^{d}$ May be interchanged. ${ }^{e}$ Coupling constant assigned by 1D TOCSY (irradiation of $\mathrm{HS}_{\mathrm{eq}}$ ).
(b)

$\theta\left(\mathrm{H3}_{e q}-\mathrm{C} 3-\mathrm{C} 9-\mathrm{H} 9_{a x}\right) \sim-90^{\circ}$
Figure 2. Selected NOESY correlations between the piperidine rings of (a) xestoproxamine A (1) and (b) xestoproxamine C (3). Points of attachment of linking chains are indicated by $\bullet$. Note, ring A has undergone ring-flip in 3 with respect to $\mathbf{1}$.
less hindered vinyl group derived from ring $B$ to give the conjugated naphthalene 14 . The UV spectrum of 14 in $\mathrm{CH}_{3} \mathrm{CN}$ showed characteristic $\lambda_{\text {max }}$ (246 and 292 nm ) for the conjugated naphthalene chromophore; however, as expected, this styrenyl system was CD silent due to the remoteness of the chromophore from the distal sphere of asymmetry near C2 (xestoproxamine numbering, Figure 1).

In order to assign the observed Cotton effect to the configuration of the C23 center in 3, an appropriate methyl-branched model was required. Because the Cotton effect anticipated for the degradation/cross metathesis product of 3 would arise from asymmetric perturbation of the vinyl-conjugated 6-methoxynaphthalene, the model could consist of a chromophore adjacent to an allylic methyl branch of known configuration, appended to a short chain. Preparation of the model was achieved as depicted in Scheme 2. The $\alpha$-branched propionamide 15, obtained by diastereoselective Myers alkylation of the enolate of $(R, R)-(-)$ -$N$-propionylpseudoephedrine ${ }^{10}$ (16) (LDA, LiCl), ${ }^{11}$ with 1-iodosilyloxy ether 17 was selectively reduced $\left(\mathrm{LiBH}_{3} \mathrm{~N}\left(\mathrm{CH}_{2}\right)_{4}\right.$, rt) to the alcohol $18^{12}$ (\% ee $>94 \%$ by modified Mosher method). ${ }^{13}$ Swern oxidation of 18 to the corresponding aldehyde followed by Wittig olefination $\left(\mathrm{Ph}_{3} \mathrm{PC}=\mathrm{CH}_{2}\right)$ gave terminal olefin 19. Finally, cross metathesis of 19 (Grubbs' II, MeO-NpV) provided model alkene (S)-20.

Xestoproxamine C (3) was converted to 21, via the intermediates 22-24 (Scheme 3), by a similar sequence of reactions used to convert 1 to $\mathbf{1 4}$. The CD spectra (Figure 3 and Table 4) of synthetic $(S)-20[\lambda 256 \mathrm{~nm}(\Delta \varepsilon+3.9)]$ and $21[\lambda 255 \mathrm{~nm}(\Delta \varepsilon$ $+3.8)]$ were of the same sign and magnitude; therefore, 21 and xestoproxamine C (3) have the $23 S$ configuration.

No simple spectroscopic method was available for defining the absolute configuration of the bis-piperidine ring system in 1-3. Since this is an outstanding problem of configurational analysis in this class of alkaloids, we turned to refinement of a general

Table 2. ${ }^{1} \mathrm{H}(600 \mathrm{MHz})$ and ${ }^{13} \mathrm{C}$ NMR $(125 \mathrm{MHz})$ for Xestoproxamine $\mathrm{B}(2)\left(\mathrm{CD}_{3} \mathrm{OD}\right)$

| no. | $\delta_{\mathrm{C}}$, mult. ${ }^{\text {a }}$ | $\delta_{\mathrm{H}}$, mult ( $J$ in Hz , ax/eq) | COSY | $\mathrm{HMBC}^{\text {b }}$ |
| :---: | :---: | :---: | :---: | :---: |
| 1 | 55.7, $\mathrm{CH}_{2}$ | 3.23, dd (12.0, 12.0, ax) | 3 | 2, 3, 5, 11, 30 |
|  |  | 3.16 , m (eq) |  | 2, 3, 5, 11, 30 |
| 2 | 29.9, CH | 2.24, m (ax) | 1,3,30 | 1, 3, 29, 30 |
| 3 | 36.1, CH | 1.87, m (ax) | 2, 4 | 1, 2, 4, 5 8, 10, 2930 |
| 4 | 25.3, $\mathrm{CH}_{2}$ | $1.81, \mathrm{~m}$ (eq) | 3, 5 | 2, 3, 911 |
|  |  | 1.68, m (ax) |  | 2, 3, 9 |
| 5 | 47.5, $\mathrm{CH}_{2}$ | 3.45 , brd (13.7, eq) | 4 | 1, 3, 4 |
|  |  | 3.17, m (ax) |  | 1, 3, 4 |
| 6 | 57.4, $\mathrm{CH}_{2}$ | $3.23, \operatorname{brd}(12.0, \mathrm{eq})$ | 7 | 7, 8, 10, 20, 21 |
|  |  | 2.85, dd (12.0, 12.0, ax) |  | 7, 8, 10, 21 |
| 7 | 34.5, CH | 1.91, m (ax) | 6, 8, 20 | 6, 8, 19, 20 |
| 8 | 27.9, $\mathrm{CH}_{2}$ | 1.87 , m (eq) | 7, 9 | 3, 6, 7, 9, 10, 20 |
|  |  | 1.08, ddd (12.0, 12.0, 12.0, ax) |  | 3, 6, 7, 9, 10, 20 |
| 9 | 42.1, CH | 1.78, m (ax) | 8, 10 | 4, 7, |
| 10 | 56.7, $\mathrm{CH}_{2}$ | $3.53, \mathrm{~m}$ (eq) | 9 | 6, 89,21 |
|  |  | 2.98, dd (12.0, 12.0, ax) |  | 6, 8, 9 |
| 11 | 49.3, $\mathrm{CH}_{2}$ | 3.30, m | 12 | 1, 5, 12, 13 |
|  |  | 3.02, m |  | 1. 5, 12, 13 |
| 12 | 21.7, $\mathrm{CH}_{2}$ | 1.77,m | 11, 13 | 11, 13, 14 |
|  |  | 1.65, m |  | 11, 13 |
| 13 | 25.4, $\mathrm{CH}_{2}$ | 1.68, m | 12, 14 | 11, 12, 14, 15 |
|  |  | 1.60, m |  | 11, 12, 15 |
| 14 | 27.5, $\mathrm{CH}_{2}$ | 1.44, m | 13, 15 | 12, 13, 15 |
| 15 | 28.9, $\mathrm{CH}_{2}$ | 1.44, m | 14, 16 | 13, 16, 17 |
|  |  | 1.26, m |  | 13, 16, 17 |
| 16 | 29.5, $\mathrm{CH}_{2}$ | 2.22, m | 15, 17 | 14, 15, 17, 18 |
|  |  | 2.00, m |  | 14, 15, 17, 18 |
| 17 | 131.7, CH | 5.48, m | 16,18 | 16, 18, 19 |
| 18 | 129.2, CH | 5.33, m | 17, 19 | 17, 18, 19 |
| 19 | 23.5, $\mathrm{CH}_{2}$ | 2.45, m | 18, 20 |  |
|  |  | 2.15, m |  | 17, 20 |
| 20 | 31.7, $\mathrm{CH}_{2}$ | 1.75, m | 19, 7 | 6, 7, 18, |
|  |  | 1.28, m |  | 6, 7, 8, 18, 19 |
| 21 | 56.1, $\mathrm{CH}_{2}$ | 3.30, m | 22 | 6, 10, 22, 23 |
|  |  | 3.07, m |  | 6, 10, 22, 23 |
| 22 | 20.0, $\mathrm{CH}_{2}$ | 1.77, m | 21, 23 | 21, 23 |
| 23 | 26.2, $\mathrm{CH}_{2}$ | 1.50, m | 22, 24 | 22 |
|  |  | 1.37, m |  | 21, 22 |
| 24-28 | 29.0-26.4, $\mathrm{CH}_{2}$ | 1.53-1.25, m |  |  |
| 29 | 21.8, $\mathrm{CH}_{2}$ | 1.25, m | 30 |  |
|  |  | 1.21, m |  |  |
| 30 | 32.1, $\mathrm{CH}_{2}$ | 1.69, m | 2, 29 | 1,2,3 |
|  |  | 1.44, m |  | 1,2,3 |

method for assignment of cyclic tertiary amines involving derivatization and CD first published by Nakanishi and coworkers. ${ }^{14}$ Nakanishi's assignment of the absolute configuration of quinuclidinols ${ }^{14}$-bicyclic tertiary amines-followed simultaneous quaternization of the tertiary nitrogen and esterification of the pendant secondary OH group with excess $p$-phenylbenzylchloride and interpretation of the resultant split CD spectrum arising from exciton coupling of the two arene chromophores. Although the xestoproxamines differ considerably from the quinuclidinols, the same principle could be applied: quaternization of both N
atoms with a suitable chromophore and interpretation of exciton coupling CD (ECCD) would be informative of the absolute configuration of the heterocyclic bis-piperidine core. However, three possible problems were anticipated. The distance between chromophores attached to the N atoms in $\mathbf{1 - 3}$ is considerably larger than those in derivatized quinuclidinols with an expectedly weaker ECCD. This could be compensated for by substitution of the $p$-phenylbenzyl group with a more polarized chromophore: a phenacyl group. It was anticipated that possible differences in conformations of the natural products and quaternized derivatives

Table 3. ${ }^{1} \mathrm{H}(600 \mathrm{MHz})$ and ${ }^{13} \mathrm{C}$ NMR ( 125 MHz$)$ NMR Data for xestoproxamine $\mathrm{C}(3)\left(\mathrm{CDCl}_{3}\right)$

| no. | $\delta_{\mathrm{C}}$, mult. ${ }^{\text {a }}$ | $\begin{gathered} \delta_{\mathrm{H}}, \text { mult } \\ (J \text { in } \mathrm{Hz}, \text { ax/eq }) \end{gathered}$ | DQF-COSY | $\mathrm{HMBC}^{\text {b }}$ |
| :---: | :---: | :---: | :---: | :---: |
| 1 | 53.1, $\mathrm{CH}_{2}$ | $\begin{aligned} & \text { 2.93, dd (12.0, } \\ & \text { 12.0, ax) } \end{aligned}$ | 2 |  |
|  |  | $2.55, \operatorname{brd}$ (12.0, eq) |  | 2, 5 |
| 2 | 40.1, CH | 2.03, m (ax) | 1, 3, 30 |  |
| 3 | 34.0, CH | 1.96, m (eq) | 2, 4 | 1, 2, 4, 8, 9, 10 |
| 4 | $34.7, \mathrm{CH}_{2}$ | 2.12, m (ax) | 3, 5 | 2 |
|  |  | 1.80, m (eq) |  |  |
| 5 | 46.7, $\mathrm{CH}_{2}$ | $\begin{aligned} & \text { 3.14, dd }(12.0, \\ & 12.0, \mathrm{ax}) \end{aligned}$ | 4 |  |
|  |  | $2.80, \mathrm{~m}$ (eq) |  |  |
| 6 | 60.0, $\mathrm{CH}_{2}$ | $2.76, \mathrm{~m}$ (eq) | 7 |  |
|  |  | 1.84, dd (12.0, |  |  |
|  |  | 12.0, ax) |  |  |
| 7 | 37.4, CH | 1.60, m (ax) | 6, 8, 20 |  |
| 8 | 37.9, $\mathrm{CH}_{2}$ | $2.43, \mathrm{~m}$ (eq) | 7, 9 | 6, 7, 9, 10, 20 |
|  |  | 1.00, ddd (12.0, |  |  |
|  |  | 12.0, |  |  |
|  |  | 12.0, ax) |  |  |
| 9 | 44.9, CH | 1.98, m (ax) | 8, 10 | 8 |
| 10 | 59.9, $\mathrm{CH}_{2}$ | 2.98, brd (9.8, eq) | 9 |  |
|  |  | 2.17, m (ax) |  | 6, 8 |
| 11 | 56.5, $\mathrm{CH}_{2}$ | 2.91, m | 12 | 1, 5, 12, 13 |
| 12 | 22.0, $\mathrm{CH}_{2}$ | 1.65, m | 11, 13 | 11, 13 |
|  |  | 1.56, m |  | 11, 13 |
| 13 | 26.8, $\mathrm{CH}_{2}$ | 1.40, m | 12 | 11, 12 |
| 14 | 26.8-27.7 | $1.50-1.20, \mathrm{~m}$ |  |  |
| 15 | 27.3, $\mathrm{CH}_{2}$ | 1.37, m |  |  |
| 16 | 26.6, $\mathrm{CH}_{2}$ | 2.03, m | 15, 17 | 15, 17, 18 |
|  |  | 1.96, m |  | 15, 17, 18 |
| 17 | 129.5, CH | 5.44, m | 16, 18 | 16 |
| 18 | 129.5, CH | 5.41, m | 17, 19 | 16 |
| 19 | 24.8, $\mathrm{CH}_{2}$ | 2.37, m | 18, 20 | 17, 18, 20 |
|  |  | 1.81, m |  | 7, 17, 18, 20 |
| v20 | 34.1, $\mathrm{CH}_{2}$ | 1.51, m | 7, 19 | 8, 18, 19 |
|  |  | 0.93, m |  | 6, 7, 8, 18, 19 |
| 21 | 55.0, $\mathrm{CH}_{2}$ | 2.70, m | 22 | 10 |
|  |  | 2.48, m |  | 6,10 |
| 22 | 27.9, $\mathrm{CH}_{2}$ | 1.58, m | 21, 23 | 21, 23, 24 |
|  |  | 1.48, m |  | 21, 24, 31 |
| 23 | 30.5, CH | 1.51, m | 22, 24, 31 |  |
| 24 | 35.9, $\mathrm{CH}_{2}$ | 1.39, m | 23, 25 | 22, 23, 31 |
|  |  | 1.09, m |  |  |
| 25-30 | 26.8-27.7 | $1.50-1.20, \mathrm{~m}$ |  |  |
| 31 | 21.2, $\mathrm{CH}_{3}$ | 0.87, d (6.7) | 23 | 22, 23, 24 |

${ }^{a}$ Assigned from HSQC/HMBC. ${ }^{b}$ HMBC correlations, optimized for 8 Hz , are from proton(s) stated to the indicated carbon.
and the introduction of two new stereocenters at N would complicate nonempirical interpretation of ECCD effects. On the other hand, the stereochemical outcome of N -quaternization should be predictable. The alkylating reagent should approach both N atoms in 1-3 along a trajectory in line with the lone pair without changing the N -configuration. In any case, the resultant
configuration and conformation of the derivatives would be revealed by CD spectra and could be interpreted as "fingerprint" spectra that relate the handedness of the heterocyclic core. This would confer an important advantage to chiroptical analysis of antipodal bis-piperidines by CD over comparisons using $[\alpha]_{\mathrm{D}}$. Optical rotations of alkanamines are typically weak in magnitude, and comparisons of specific rotation alone are notoriously unreliable due to changes in sign resulting from even slight structural variants (unsaturation). By comparing a fingerprint Cotton effect with that of a bis-piperidine phenacyl derivative of known absolute configuration, the configuration of any member of the series could be assigned.

To ensure a reasonably strong CD signal in the final product, an $N$-(4-bromophenacyl) chromophore ( $\lambda_{\max } 265 \mathrm{~nm}, \varepsilon$ $\sim 12000$ ) was introduced by exhaustive quaternization of the alkaloids. The bis-TFA salt of tetrahydroxestoproxamine (10) was converted to the free base $(\mathrm{KOH} / \mathrm{MeOH})$ followed by alkylation with $p$-bromophenacyl bromide (toluene, $90^{\circ} \mathrm{C}$ ) to give $\mathbf{2 5}$ after HPLC purification (Scheme 4). The CD spectrum (Table 4) of 25 showed a characteristic negative split Cotton effect ( $\lambda 273 \mathrm{~nm}, \Delta \varepsilon-6.0 ; 253(+2.6)$ ) arising from exciton coupling of equatorial and axial $N$-p-bromophenacyl chromophores disposed with a negative helicity (see Figure 4). The relative conformations of the conjoined bis-piperidine ring system in 25 and $\mathbf{1}$ maintain the same conformations as starting material (analysis of $J$ and NOESY data); however, the exact orientation of the transition dipole vectors that align along the average axes of the $N-p-\mathrm{Br}$-phenacyl groups is highly dependent upon the conjoining $\mathrm{C} 3-\mathrm{C} 9$ bond, which is subject to subtle torsional angle changes from nonbonding effects of the side chains and not easily predictable. Instead, we chose to make empirical chiroptical comparisons with $p$-Br-phenacyl derivatives with a compound of known configuration.

The relative configuration of $(-)$-"perhaliclonacyclamine" (26), ${ }^{15}$ the hydrogenation product of $(+)$-tetradehydrohaliclonacyclamine A (31) obtained from an Indonesian sponge, Halichondria sp., differs from the tetrahydroxestoproxamine A (10) in the relative configuration at C 2 and the presence of two additional $\mathrm{CH}_{2}$ groups in the lower linking chain. ${ }^{16}$ A sample of $(-)-26^{17}$ was alkylated in the same manner (see above) to give derivative 27. The CD spectrum of 27 ( $\lambda 273 \mathrm{~nm}(\Delta \varepsilon-6.1)$; $254,(+3.2)$; Figure 4, Table 4) is almost identical to that of 25. Therefore, despite differences in relative configuration in the two piperidine rings in 25 and 27, both conform to the same chromophore alignments (a negative $\mathrm{N}-(\mathrm{C}=\mathrm{O})-(\mathrm{C}=\mathrm{O})-\mathrm{N}^{\prime}$ angle) -imposed largely by the constraint of the C3-C9 torsional angle-and CD reveals they share the same absolute stereostructure form with the exception of the epimeric C 2 position. A sample of haliclonacyclamine E (9), isolated from a Brazilian sample of Arenosclera braziliensis, ${ }^{18}$ was hydrogenated ( $\mathrm{H}_{2}, \mathrm{Pd}-$ C) to the tetrahydro derivative (28) and exhaustively alkylated ( $p$ bromophenacyl bromide, $90{ }^{\circ} \mathrm{C}$ ) as before to the bis- $p$ - Br phenacylated compound 29, which showed a CD spectrum ( $\lambda$ $273 \mathrm{~nm}, \Delta \varepsilon-6.0 ; 255(+2.6)$, Table 4) almost identical with those of 25 and 27. Finally, the CD spectrum of the bis-pbromophenylacyl derivative ( $30, \lambda 273 \mathrm{~nm}, \Delta \varepsilon-7.1 ; 253$ $(+3.5)$ ), obtained from tetrahydroxestoproxamine C (22), was identical to those of 25,27 , and $30 .{ }^{19}$ Therefore the absolute configuration of the heterocyclic cores in xestoproxamine A (1), haliclonacyclamine E (9), ( - )-perhaliclonacyclamine (26), and xestoproxamine $C$ (3) are the same, with the exception of C 2 as noted above. Since the absolute configuration of $(-)-26$ was

Scheme 1. Hofmann Degradation of Xestoproxamine A (1) and Cross Metathesis with 6-MeO-2-vinylnaphthalene (13)




1) Grubbs II cat.


12

14


Scheme 2. Synthesis of Model Compound 20



Scheme 3. Hofmann Degradation of Xestoproxamine C (3) and Cross Metathesis with 6-MeO-2-vinylnaphthalene (13)

established earlier by X-ray crystallography, ${ }^{15}$ the stereostructures of $1-3$ and 9 are now completely assigned. Bis-piperidine 8 cooccurs with $9,{ }^{7}$ and it is likely both of these natural products share the same heterocyclic core configuration.

A summary of the configurational analysis of 1-3 and the CIP descriptors for the core stereocenters of related bis-piperidine
alkaloids is given in Table 5, along with a comparison of reported $[\alpha]_{\mathrm{D}}$ values. No clear trend can be seen except that the compounds have high stereochemical heterogeneity that may reflect their geographic origins. The specific rotations vary in sign and magnitude with a strong dependence upon the presence and position of unsaturation in the top and bottom linking chains and possibly


Figure 3. CD spectra $\left(\mathrm{CH}_{3} \mathrm{CN}, 23{ }^{\circ} \mathrm{C}\right)$ of (a) 20, (b) 21, and (c) 14.
Scheme 4. Preparation of $N, N^{\prime}-p-B r-p h e n a c y l ~ D e r i v a t i v e s ~ o f ~$ Bis-piperidine Alkaloids

the chain lengths. Haliclonacyclamines A (6) and B (7), from Haliclona sp. from the Great Barrier Reef, differ in the linking chains, but the cores are antipodal to those of $(-)-26$ and $(+)-31$ from Halichondria sp. collected in Indonesia. We note that the recently reported $(-)$-neopetrosiamine $(32)^{20}$ from the sponge Neopetrosia proxima, ${ }^{2}$ collected in the Caribbean sea south of the Bahamas, is very similar to $\mathbf{1 - 3}$, but the configuration was not defined.

Bis-piperidine alkaloids have shown modest activity against various cancer cell lines. Compounds $\mathbf{1 - 3}$ were assayed in vitro against human colon tumor cells (HCT-116) and showed $\mathrm{IC}_{50}$ values of $21.2,6.3$, and $5.4 \mu \mathrm{M}$, respectively.

In conclusion, three new bis-piperidine alkaloids, xestoproxamines $\mathrm{A}-\mathrm{C}(1-3)$, are reported and their complete structures assigned by integrated MS, NMR, and chiroptical analysis. We assigned the absolute configuration of the remote methyl group in xestoproxamine $\mathrm{C}(3)$ by CD following a Hoffman degradation/cross metathesis protocol, a sensitive technique that can be extended to other natural products containing acyclic allylic methyl branches. Finally, we have shown that ECCD of $N, N^{\prime}$-di-p-bromophenacyl bis-piperidine alkaloids, obtained by

Table 4. CD Data ( $23{ }^{\circ} \mathrm{C}$ ) for Vinylnaphthalene and Bis-pbromophenacyl Derivatives

| no. | solvent | $\lambda / \mathrm{nm}$ | $\Delta \varepsilon^{b}$ | $\lambda / \mathrm{nm}$ | $\Delta \varepsilon^{b}$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathbf{1 4}^{c}$ | $\mathrm{CH}_{3} \mathrm{CN}$ |  |  |  | $a$ |
| $\mathbf{2 0}^{d}$ | $\mathrm{CH}_{3} \mathrm{CN}$ |  |  | 255 | +3.8 |
| $\mathbf{2 1}^{e}$ | $\mathrm{CH}_{3} \mathrm{CN}$ |  |  | 256 | +3.9 |
| $\mathbf{2 5}$ | $\mathrm{CH}_{3} \mathrm{OH}$ | 276 | -6.2 | 254 | +2.6 |
| $\mathbf{2 7}^{f}$ | $\mathrm{CH}_{3} \mathrm{OH}$ | 272 | -6.9 | 254 | +3.2 |
| $\mathbf{2 9}$ | $\mathrm{CH}_{3} \mathrm{OH}$ | 272 | -6.0 | 254 | +2.6 |
| $\mathbf{3 0 ^ { f }}$ | $\mathrm{CH}_{3} \mathrm{OH}$ | 276 | -7.1 | 254 | +3.5 |

${ }^{a}$ Only noise. ${ }^{b} \mathrm{~mol}^{-1} \mathrm{dm}^{3} \mathrm{~cm}^{-1} .{ }^{c}$ Scheme $1 .{ }^{d}$ Scheme $2 .{ }^{e}$ Scheme 3.
${ }^{f}$ Scheme 4.


Figure 4. CD spectra $\left(\mathrm{MeOH}, 23{ }^{\circ} \mathrm{C}\right)$ of (a) 25, (b) 27, (c) 29, and (d) 30 .
quaternization of the tertiary amines, reliably reflects the absolute configurations of the heterocyclic cores in $1-3$, independent of the relative configuration at C 2 . This observation was exploited to show the two groups of structures, $\mathbf{1 , 2 , 8 , 9}$, and 3,26 , have the same absolute configuration in their core heterocyclic rings and should be applicable to other bis-piperidine alkaloids in this class.

## ■ EXPERIMENTAL SECTION

General Experimental Procedures. General procedures are described elsewhere. ${ }^{21}$ Yields were determined gravimetrically except for those of masses of less than $\sim 100 \mu \mathrm{~g}$, which are estimated from UV extinction coefficients or by quantitative solvent ${ }^{13} \mathrm{C}$ satellite (QSCS) ${ }^{22}$ analysis of cryomicroprobe-measured ${ }^{1} \mathrm{H}$ NMR spectra. HPLC was carried out using either a Gilson model 302 pump equipped with tandem detectors-UV-visible (ISCO model UA-5, $\lambda 254 \mathrm{~nm}$ ) and refractive index (Waters R401)—or a Rainin HPXL dual-pump with split flow (7:1) between two detectors-a Jasco CD-2095 UV-CD and an ESA model 301 evaporative light scattering detector (ELSD).

Animal Material. The sponge Neopetrosia proxima ${ }^{2}$ was collected in June 2008 at Stirrup Cay, the Bahamas $\left(25^{\circ} 49.511 \mathrm{~N} 77^{\circ} 53.924^{\prime} \mathrm{W}\right)$ at a depth of 8.5 m and identified by Sven Zea (Universidad Nacional de Colombia, InveMar). The surface of the tissue was dark brown and free of epiphytic zoanthids. A voucher sample of the sponge (08-13-073) is stored at UC San Diego.

Table 5. Specific Rotations and Configurational Assignment of Bis-piperidine Alkaloids ${ }^{a}$

| cmpd | $[\alpha]_{\mathrm{D}}$ | conc, g/ 100 mL | solvent | linker chain $\mathrm{C}_{n}$ |  | absolute confign ${ }^{a}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  | top | bottom |  |
| $1^{b}$ | +4.4 | c 2.0 | $\mathrm{CH}_{3} \mathrm{OH}$ | $\mathrm{C}_{10}$ | $\mathrm{C}_{10}$ | 2R,3S,7S, 9 S |
| $2^{\text {b }}$ | +2.7 | c 2.4 | $\mathrm{CH}_{3} \mathrm{OH}$ | $\mathrm{C}_{10}$ | $\mathrm{C}_{10}$ | 2R,3S,7S, 9 S |
| $3^{b}$ | -18.5 | c 0.67 | $\mathrm{CHCl}_{3}$ | $\mathrm{C}_{10}$ | $\mathrm{C}_{10}$ | 2S,3S,7S,9S,23S |
| $4{ }^{\text {c }}$ | -7.3 | c 0.73 | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | $\mathrm{C}_{8}$ | $\mathrm{C}_{12}$ | $2 R^{*}, 3 S^{*}, 9 R^{*} h$ |
| $5^{d}$ | -143.5 | c 0.65 | ? | $\mathrm{C}_{10}$ | $\mathrm{C}_{8}$ | $2 R^{*}, 3 S^{*}, 7 S^{* h}$ |
| $6^{\text {e }}$ | -3.4 | c 1.21 | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | $\mathrm{C}_{10}$ | $\mathrm{C}_{12}$ | $2 R, 3 R, 7 R, 9 R$ |
| $7{ }^{\text {e }}$ | +3.4 | c 0.55 | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | $\mathrm{C}_{10}$ | $\mathrm{C}_{12}$ | 2R,3R,7R,9R |
| $8^{f}$ | -3 | c 0.015 | MeOH | $\mathrm{C}_{10}$ | $\mathrm{C}_{12}$ | $2 R, 3 S, 7 S, 9 S^{\text {b }}$ |
| $9^{f}$ | +14 | c 0.02 | MeOH | $\mathrm{C}_{10}$ | $\mathrm{C}_{12}$ | 2R,3S,7S, $9 S^{\text {b }}$ |
| $26^{g}$ | -20.9 | c 0.205 | $\mathrm{CHCl}_{3}$ | $\mathrm{C}_{10}$ | $\mathrm{C}_{12}$ | 2S,3S,7S,9S |
| $31^{g}$ | +19.4 | c 0.515 | $\mathrm{CHCl}_{3}$ | $\mathrm{C}_{10}$ | $\mathrm{C}_{12}$ | 2S,3S,7S,9S |
| $32^{g}$ | -10 | c 1.0 | $\mathrm{CHCl}_{3}$ | $\mathrm{C}_{10}$ | $\mathrm{C}_{10}$ | $2 R^{*}, 3 R^{*}, 7 R^{*}, 9 R^{* h}$ |

${ }^{a}$ Core locants follow the numbering used for $\mathbf{1 - 3}$. Unless indicated by ${ }^{*}$, depicted CIP assignments are absolute. Configuration of the N atoms not given. ${ }^{b}$ This work. ${ }^{c}$ Ref 4. ${ }^{d}$ X-ray, ref 5. ${ }^{e}$ X-ray, ref 6a. ${ }^{f}$ ref 7. ${ }^{g}$ X-ray, ref $15 .{ }^{h}$ Relative configuration only. ${ }^{g}$ Ref 20.

Extraction and Isolation. A sample of $N$. proxima ( 252.6 g wet wt.) was extracted with $\mathrm{MeOH}\left(3 \times 2.5 \mathrm{~L}, 25^{\circ} \mathrm{C}\right.$, overnight) and then $\mathrm{CH}_{2} \mathrm{Cl}_{2} /$ $\mathrm{MeOH}\left(2 \times 2.5 \mathrm{~L}, 25^{\circ} \mathrm{C}\right.$, overnight). The resulting extract was partitioned between hexanes $(3 \times 500 \mathrm{~mL})$ and 9:1 $\mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}(1 \mathrm{~L})$. The aqueous MeOH layer was removed and the $\mathrm{H}_{2} \mathrm{O}$ content adjusted to $1: 1 \mathrm{MeOH} /$ $\mathrm{H}_{2} \mathrm{O}$ before extraction with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 1 \mathrm{~L})$. The aqueous MeOH layer was concentrated under reduced pressure and then applied directly onto a reversed-phase $\left(\mathrm{C}_{18}\right)$ silica column $(\sim 400 \mathrm{~g})$ successively eluting with 1:9, $3: 7,1: 1$, and $9: 1 \mathrm{CH}_{3} \mathrm{CN} / \mathrm{H}_{2} \mathrm{O}+0.2 \%$ TFA, and $i$-PrOH $+0.2 \%$ TFA to give five fractions. A portion $(305 \mathrm{mg})$ of the third fraction $(1.22 \mathrm{~g})$ was subjected to preparative HPLC (reversed-phase, $\mathrm{C}_{18}, 10 \mathrm{~mL} \mathrm{~min}^{-1}$, gradient elution with $3: 7$ to $2: 5 \mathrm{CH}_{3} \mathrm{CN} / \mathrm{H}_{2} \mathrm{O}+0.3 \%$ TFA over 40 min ) to give 12 fractions. The third preparative HPLC fraction $(27.2 \mathrm{mg})$ was subjected to semipreparative HPLC (reversed-phase Synergi-HydroRP, $2.5 \mathrm{~mL} \mathrm{~min}{ }^{-1}$, mobile phase: 2:2:6:0.05 $\mathrm{CH}_{3} \mathrm{CN} / i$-PrOH/ $\left.\mathrm{H}_{2} \mathrm{O} / \mathrm{TFA}\right)$ to give xestoproxamine $A(1,14.0 \mathrm{mg})$. A portion $(4 \mathrm{mg})$ of the fourth preparative HPLC fraction $(24.0 \mathrm{mg})$ was subjected to semipreparative HPLC over the same column ( $2.5 \mathrm{~mL} \mathrm{~min}^{-1}$, 22.5:22.5:55:0.5 $\mathrm{CH}_{3} \mathrm{CN} / i-\mathrm{PrOH} / \mathrm{H}_{2} \mathrm{O} / \mathrm{TFA}$ ) to give xestoproxamine $B(2,2.9 \mathrm{mg})$. The entire sixth HPLC fraction was subjected to HPLC (Phenomenex Lux-cellulose, $1.5 \mathrm{~mL} . \mathrm{min}^{-1}$, mobile phase: 9:1:0.02 $\left.\mathrm{CH}_{3} \mathrm{CN} / i-\mathrm{PrOH} / \mathrm{Et}_{2} \mathrm{NH}\right)$ to give xestoproxamine $\mathrm{C}(3,1.5 \mathrm{mg})$.

Xestoproxamine $A(1):$ : colorless glass; $[\alpha]_{\mathrm{D}}^{24}+4.4(c 2.0, \mathrm{MeOH})$; FTIR (ATR) $v 1676,1437,1203,1133,841,801,723 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see Table 1; HRESIMS $m / z 441.4207[\mathrm{M}+\mathrm{H}]^{+}$(calcd for $\mathrm{C}_{30} \mathrm{H}_{53} \mathrm{~N}_{2}, 441.4203$ ).

Xestoproxamine $B(\mathbf{2}):$. colorless glass; $[\alpha]_{\mathrm{D}}^{24}+2.7(c 2.4, \mathrm{MeOH})$; FTIR (ATR) $v$ 2934, 2862, 1674, 1463, 1199, 1131, 833, 799, $721 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see Table 2; HRESIMS $m / z 443.4362[\mathrm{M}+\mathrm{H}]^{+}$ (calcd for $\mathrm{C}_{30} \mathrm{H}_{55} \mathrm{~N}_{2}, 443.4358$ ).

Xestoproxamine C (3):: colorless solid; $[\alpha]^{24}{ }_{\mathrm{D}}-18.5$ (c 0.67, $\mathrm{CHCl}_{3}$ ); FTIR (ATR) $v 2933,2862,1673,1470,1199,1128,829$, $797,721 \mathrm{~cm}^{-1}$; $^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see Table 3; HRESIMS $\mathrm{m} / z 457.4513$ $[\mathrm{M}+\mathrm{H}]^{+}\left(\right.$calcd for $\left.\mathrm{C}_{31} \mathrm{H}_{57} \mathrm{~N}_{2}, 457.4516\right)$.

Tetrahydroxestoproxamine A Free Base (10). A solution of xestoproxamine A (1) TFA salt ( 1.0 mg ) and $\mathrm{Pd} / \mathrm{C}(10 \%, 0.2 \mathrm{mg})$ in MeOH $+2 \% \mathrm{AcOH}(0.5 \mathrm{~mL})$ was vigorously stirred under 1 atm of $\mathrm{H}_{2}$ overnight. The mixture was passed through a membrane filter ( 0.45 $\mu \mathrm{m})$. The solvent was removed from the filtrate under reduced pressure,
and the residue subjected to HPLC (Phenomenex Lux-cellulose, 1.5 mL . $\min ^{-1}$, mobile phase: 9:1:0.02 $\left.\mathrm{CH}_{3} \mathrm{CN} / i-\mathrm{PrOH} / \mathrm{Et}_{2} \mathrm{NH}\right)$ to give tetrahydroxestoproxamine A free base $(\mathbf{1 0}, 0.55 \mathrm{mg})$, which was used immediately after characterization. Colorless solid; $[\alpha]^{23}{ }_{\mathrm{D}}-12.1$ (c $\left.0.7, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 2.90(\mathrm{dt}, 1 \mathrm{H}, \mathrm{J}=12.0,5.5$ $\mathrm{Hz}), 2.80(\mathrm{dd}, 1 \mathrm{H}, J=12.5,4.0 \mathrm{~Hz}), 2.73(\mathrm{brd}, 1 \mathrm{H}, J=10.0 \mathrm{~Hz}), 2.70-$ $2.61(\mathrm{~m}, 5 \mathrm{H}), 2.54(\mathrm{brp}, 1 \mathrm{H}, J=5.5 \mathrm{~Hz}), 2.41($ brt, $1 \mathrm{H}, J=9.8 \mathrm{~Hz}), 2.24$ $(\mathrm{t}, 1 \mathrm{H}, 11 \mathrm{~Hz}), 2.04(\mathrm{~d}, 1 \mathrm{H}, J=12.5 \mathrm{~Hz}), 1.87(\mathrm{t}, 1 \mathrm{H}, J=11 \mathrm{~Hz}), 1.82(\mathrm{~m}$, $1 \mathrm{H}), 1.64-1.21(\mathrm{~m}, 40 \mathrm{H}), 0.97(\mathrm{~m}, 1 \mathrm{H}), 0.72(\mathrm{q}, 1 \mathrm{H}, J=12 \mathrm{~Hz})$; HRESIMS $m / z 445.4518[\mathrm{M}+\mathrm{H}]^{+}\left(\right.$calcd for $\left.\mathrm{C}_{30} \mathrm{H}_{57} \mathrm{~N}_{2}, 445.4516\right)$.

Tetrahydroxestoproxamine A Bis-methiodide Salt (11). Free base $10(0.55 \mathrm{mg})$ was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.8 \mathrm{~mL})$, excess $\mathrm{CH}_{3} \mathrm{I}(0.2 \mathrm{~mL})$ was added, and the mixture was stirred overnight in the dark. Volatiles were removed under a stream of $\mathrm{N}_{2}$ to give the bis-methiodide salt $(\mathbf{1 1}, 0.9 \mathrm{mg})$ as an off-white solid: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 3.84(\mathrm{~d}, J=10.5 \mathrm{~Hz}$, $1 \mathrm{H}), 3.54-3.49(\mathrm{~m}, 3 \mathrm{H}), 3.41-3.15(\mathrm{~m}, 8 \mathrm{H}), 3.17(\mathrm{~s}, 3 \mathrm{H}), 3.12(\mathrm{~s}, 3 \mathrm{H})$, $2.20-1.22(\mathrm{~m}, 44 \mathrm{H}), 1.19(\mathrm{q}, J=12.5 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $(125 \mathrm{MHz}$, $\left.\mathrm{CD}_{3} \mathrm{OD}\right) \delta 69.4\left(\mathrm{CH}_{2}\right), 66.5\left(\mathrm{CH}_{2}\right), 64.6\left(\mathrm{CH}_{2}\right), 63.6\left(\mathrm{CH}_{2}\right), 60.4$ $\left(\mathrm{CH}_{2}\right), 57.9\left(\mathrm{CH}_{2}\right), 53.3\left(\mathrm{CH}_{3}\right), 47.6\left(\mathrm{CH}_{3}\right), 37.0(\mathrm{CH}), 35.2(\mathrm{CH}), 32.2$ $(\mathrm{CH}), 32.1\left(\mathrm{CH}_{2}\right), 31.8\left(\mathrm{CH}_{2}\right), 31.7(\mathrm{CH}), 29.2\left(\mathrm{CH}_{2}\right), 29.1\left(\mathrm{CH}_{2}\right), 29.0$ $\left(\mathrm{CH}_{2}\right), 28.8\left(\mathrm{CH}_{2}\right), 28.6\left(2 \times \mathrm{CH}_{2}\right), 28.3\left(\mathrm{CH}_{2}\right), 28.2\left(\mathrm{CH}_{2}\right), 28.1$ $\left(\mathrm{CH}_{2}\right), 27.4\left(\mathrm{CH}_{2}\right), 27.3\left(\mathrm{CH}_{2}\right), 26.8\left(\mathrm{CH}_{2}\right), 26.6\left(\mathrm{CH}_{2}\right), 25.7\left(\mathrm{CH}_{2}\right)$, $25.3\left(\mathrm{CH}_{2}\right), 22.3\left(\mathrm{CH}_{2}\right), 20.7\left(\mathrm{CH}_{2}\right), 20.2\left(\mathrm{CH}_{2}\right)$; HRESIMS $\mathrm{m} / \mathrm{z}$ $237.2453[\mathrm{M}]^{2+}$ (calcd for $\mathrm{C}_{32} \mathrm{H}_{62} \mathrm{~N}_{2}, 474.4908$ )

Hofmann Degradation of Tetrahydroxestoproxamine A Bismethiodide Salt. A solution of the bis-methiodide (11, 0.9 mg ) was eluted with MeOH through a short column of strong anion exchange resin (Amberlite IRA-400, $\mathrm{HO}^{-}$form, prepared immediately before from $\mathrm{Cl}^{-}$form by elution with aqueous 1 N NaOH , followed by washing with distilled $\mathrm{H}_{2} \mathrm{O}$ until the eluate was neutral, then MeOH$)$. The eluate was concentrated under reduced pressure to remove the volatiles, and solvent-free methohydroxide was irradiated (microwave, $300 \mathrm{~W}, 140^{\circ} \mathrm{C}$, 10 min ). The crude product was taken up in MeOH and passed through a reversed-phase silica $\left(\mathrm{C}_{18}\right)$ cartridge $(200 \mathrm{mg})$ by elution with MeOH $+0.1 \%$ TFA. The solvent was removed under a stream of $\mathrm{N}_{2}$, and the residue subjected to HPLC (reversed-phase, $\mathrm{C}_{18}$ Phenomenex Luna, 5 $\mu \mathrm{m}, 10 \times 250 \mathrm{~mm}$, gradient: $40-100 \% \mathrm{CH}_{3} \mathrm{CN} / \mathrm{H}_{2} \mathrm{O}+0.1 \%$ TFA over 40 min ) to give a single major elimination product, $12(0.45 \mathrm{mg})$.

12: ${ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right.$, representative signals) $\delta 5.80$ (dddd, $J=17.0,10.3,6.8,6.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.55(\mathrm{dt}, J=17.0,10.3 \mathrm{~Hz}, 1 \mathrm{H}), 5.31(\mathrm{dd}, J=$ $10.3,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.18(\mathrm{dd}, J=17.0,1.4 \mathrm{~Hz}), 4.98(\mathrm{dq}, J=17.0,1.9 \mathrm{HZ}$, $1 \mathrm{H}), 4.92(1 \mathrm{H}$, under solvent), 3.44 (brd, $J=12.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.40$ (brd, $J=12.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.20(\mathrm{td}, J=12.5,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.10(\mathrm{td}, J=12.5$, $5.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.92(\mathrm{~s}, 3 \mathrm{H}), 2.85(\mathrm{~s}, 3 \mathrm{H}), 2.54(\mathrm{t}, J=12 \mathrm{~Hz}, 1 \mathrm{H}), 2.53(\mathrm{t}, J=$ $12 \mathrm{~Hz}, 1 \mathrm{H}), 2.18(\mathrm{t}, J=10.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.05(\mathrm{q}, 7.3 \mathrm{~Hz}, 2 \mathrm{H}), 0.88(\mathrm{q}, J=12$ $\mathrm{Hz}, 1 \mathrm{H})$; HRESIMS $m / z 473.4828[\mathrm{M}+\mathrm{H}]^{+}\left(\right.$calcd for $\mathrm{C}_{32} \mathrm{H}_{61} \mathrm{~N}_{2}$, 473.4829).

Cross Metathesis of $\mathbf{1 2}$ with 2-Methoxy-6-vinylnaphthalene (13). To a solution of alkene $12(450 \mu \mathrm{~g}, 0.642 \mu \mathrm{~mol})$ and 6-meth-oxy-2-vinylnaphthalene ${ }^{8}(13,1.2 \mathrm{mg}, 6.42 \mu \mathrm{~mol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.5 \mathrm{~mL})$ was added Grubbs' second-generation catalyst ${ }^{9}(226 \mu \mathrm{~g}, 0.265 \mu \mathrm{~mol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1 \mathrm{~mL})$ in three portions over a period of 11 h at $80^{\circ} \mathrm{C}$. The solvent was evaporated under a stream of $\mathrm{N}_{2}$. The crude reaction mixture was passed through a reversed-phase $\mathrm{C}_{18}$ cartridge by elution with $\mathrm{MeOH}+0.1 \%$ TFA, then subjected to HPLC (Luna, $\mathrm{C}_{18}, 250 \times$ $10 \mathrm{~mm}, 40-100 \% \mathrm{CH}_{3} \mathrm{CN} / \mathrm{H}_{2} \mathrm{O}+0.1 \%$ TFA over 40 min ) to give the 6-methoxy-2-naphthylethenyl derivative $14(240 \mu \mathrm{~g})$ : UV $\left(\mathrm{CH}_{3} \mathrm{CN}\right)$ $\lambda_{\text {max }} 246,292 \mathrm{~nm}$; CD see Table 4; ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$, representative signals) $\delta 7.685(\mathrm{~d}, J=9.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.681(\mathrm{~d}, J=8.6 \mathrm{~Hz}$, $1 \mathrm{H}), 7.60$ (brs, 1 H ), 7.55 (dd, $J=8.6,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.18(\mathrm{~d}, J=2.0 \mathrm{~Hz}$, $1 \mathrm{H}), 7.09(\mathrm{dd}, J=9.0,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.51(\mathrm{~d}, J=16.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.32(\mathrm{dt}, J=$ $16.0,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.55(\mathrm{dt}, J=17.0,10.3 \mathrm{~Hz}, 1 \mathrm{H}), 5.31(\mathrm{dd}, J=10.3,1.4$ $\mathrm{Hz}, 1 \mathrm{H}), 5.18(\mathrm{dd}, J=17.0,1.4 \mathrm{~Hz}), 3.90(\mathrm{~s}, 3 \mathrm{H}), 2.91(\mathrm{~s}, 3 \mathrm{H}), 2.83(\mathrm{~s}$, $3 \mathrm{H}), 2.50(\mathrm{t}, J=12 \mathrm{~Hz}, 1 \mathrm{H}), 2.49(\mathrm{t}, J=12 \mathrm{~Hz}, 1 \mathrm{H}), 2.26(\mathrm{q}, J=7.0 \mathrm{~Hz}$,
$2 \mathrm{H}), 0.86(\mathrm{q}, J=12 \mathrm{~Hz}, 1 \mathrm{H})$; HRESIMS $m / z 629.5403[\mathrm{M}]^{+}$(calcd for $\left.\mathrm{C}_{43} \mathrm{H}_{69} \mathrm{~N}_{2} \mathrm{O}, 629.5404\right)$.

Degradation of Xestoproxamine C. Dihydroxestoproxamine C TFA Salt (22). A solution of xestoproxamine C (3) ( 0.8 mg ) and $\mathrm{Pd} /$ $\mathrm{C}(10 \%, 0.2 \mathrm{mg})$ in $\mathrm{MeOH}+2 \% \mathrm{AcOH}(0.5 \mathrm{~mL})$ was vigorously stirred overnight under $\mathrm{H}_{2}(1 \mathrm{~atm})$. The solution was passed through a membrane filter ( 0.45 m ), and the volatiles were removed under a stream of $\mathrm{N}_{2}$. The residue was purified by HPLC (reversed-phase, Phenomenex, Synergi-HydroRP, $4 \mu \mathrm{~m}, 10 \times 250 \mathrm{~mm} ; 3 \mathrm{~mL} \mathrm{~min}^{-1}$; mobile phase: 25:25:50:0.5 $\left.\mathrm{CH}_{3} \mathrm{CN} / i-\mathrm{PrOH} / \mathrm{H}_{2} \mathrm{O} / \mathrm{TFA}\right)$ to give dihydroxestoproxamine C (22): colorless glass; ${ }^{1} \mathrm{H}$ NMR ( 500 MHz , $\mathrm{CD}_{3} \mathrm{OD}$, representative signals) $\delta 3.68$ (brd, $\left.J=11.2 \mathrm{~Hz}, 1 \mathrm{H}\right), 3.48$ $(\mathrm{td}, J=13.0,4.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.91(\mathrm{t}, J=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.84(\mathrm{t}, J=12.0 \mathrm{~Hz}$, $1 \mathrm{H}), 2.33(\operatorname{brd}, J=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 0.95(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(125 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 58.0\left(\mathrm{CH}_{2}\right), 57.5\left(\mathrm{CH}_{2}\right), 56.6\left(\mathrm{CH}_{2}\right), 54.9$ $\left(\mathrm{CH}_{2}\right), 53.1\left(\mathrm{CH}_{2}\right), 43.3(\mathrm{CH}), 40.3(\mathrm{CH}), 37.1(\mathrm{CH}), 35.74\left(\mathrm{CH}_{2}\right)$, $35.67\left(\mathrm{CH}_{2}\right), 33.7\left(\mathrm{CH}_{2}\right), 33.6\left(\mathrm{CH}_{2}\right), 33.3(\mathrm{CH}), 32.9\left(\mathrm{CH}_{2}\right), 31.5$ $(\mathrm{CH}), 28.8\left(\mathrm{CH}_{2}\right), 25.56\left(\mathrm{CH}_{2}\right), 28.53\left(\mathrm{CH}_{2}\right), 27.8\left(3 \times \mathrm{CH}_{2}\right), 27.7$ $\left(\mathrm{CH}_{2}\right), 26.8\left(\mathrm{CH}_{2}\right), 26.6\left(\mathrm{CH}_{2}\right), 26.3\left(\mathrm{CH}_{2}\right), 26.2\left(\mathrm{CH}_{2}\right), 26.0\left(\mathrm{CH}_{2}\right)$, $25.6\left(\mathrm{CH}_{2}\right), 21.9\left(\mathrm{CH}_{2}\right), 21.0\left(\mathrm{CH}_{3}\right)$; HRESIMS $m / z 460.4748[\mathrm{M}+$ $\mathrm{H}]^{+}\left(\right.$calcd for $\left.\mathrm{C}_{31} \mathrm{H}_{60} \mathrm{~N}_{2}, 460.4751\right)$.

Tetrahydroxestoproxamine C Bis-methiodide Salt (23). The TFA salt 22 was subjected to HPLC (Lux-cellulose, $1.5 \mathrm{~mL} \mathrm{~min}{ }^{-1}$, mobile phase: 9:1:0.02 $\left.\mathrm{CH}_{3} \mathrm{CN} / i-\mathrm{PrOH} / \mathrm{DEA}\right)$ to give the free base $(0.5 \mathrm{mg})$. Immediately after removal of the volatiles, the free base $(0.5 \mathrm{mg})$ was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.8 \mathrm{~mL})$ and treated with excess $\mathrm{CH}_{3} \mathrm{I}(0.2 \mathrm{~mL})$, and the mixture was stirred overnight in the dark. The solvent was evaporated under a stream of $\mathrm{N}_{2}$ to give the methiodide $(23,0.8 \mathrm{mg})$ as an off-white solid: ${ }^{1} \mathrm{H} \mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right.$, representative signals) $\delta 3.87$ (brd, $J=11.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.77(\mathrm{td}, J=13.8,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.21$ (s, $3 \mathrm{H}), 3.12(\mathrm{~s}, 3 \mathrm{H}), 3.04(\mathrm{t}, J=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.03(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.125 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 69.5\left(\mathrm{CH}_{2}\right), 69.0\left(\mathrm{CH}_{2}\right), 67.4\left(\mathrm{CH}_{2}\right)$, $65.3\left(\mathrm{CH}_{2}\right), 61.7\left(\mathrm{CH}_{2}\right), 57.2\left(\mathrm{CH}_{2}\right)$ [obscured by solvent] $\left(\mathrm{CH}_{3}\right), 45.9$ $\left(\mathrm{CH}_{3}\right), 39.3(\mathrm{CH}), 36.4\left(\mathrm{CH}_{2}\right), 36.2(\mathrm{CH}), 35.2\left(\mathrm{CH}_{2}\right), 34.0\left(\mathrm{CH}_{2}\right)$, $33.3(\mathrm{CH}), 33.1(\mathrm{CH}), 32.6\left(\mathrm{CH}_{2}\right), 31.2(\mathrm{CH}), 30.2\left(\mathrm{CH}_{2}\right), 29.5$ $\left(\mathrm{CH}_{2}\right), 29.3\left(\mathrm{CH}_{2}\right), 28.7\left(\mathrm{CH}_{2}\right), 28.4\left(\mathrm{CH}_{2}\right), 28.2\left(\mathrm{CH}_{2}\right), 27.94\left(\mathrm{CH}_{2}\right)$, $27.86\left(2 \times \mathrm{CH}_{2}\right), 27.5\left(\mathrm{CH}_{2}\right), 26.9\left(\mathrm{CH}_{2}\right), 26.1\left(\mathrm{CH}_{2}\right), 25.3\left(\mathrm{CH}_{2}\right)$, $25.2\left(\mathrm{CH}_{2}\right), 22.1\left(\mathrm{CH}_{3}\right), 22.0\left(\mathrm{CH}_{2}\right)$; HRESIMS $m / z 244.2531[\mathrm{M}]^{2+}$ (calcd for $\mathrm{C}_{33} \mathrm{H}_{64} \mathrm{~N}_{2}, 488.5064$ ).

Hofmann Degradation of Tetrahydroxestoproxamine C Bismethiodide Salt (23). A solution of the methiodide 23 in MeOH was passed through a short column of strong anion exchange resin (Amberlite IRA-400 $\left(\mathrm{Cl}^{-}\right)$, converted to the $\mathrm{HO}^{-}$form with 1 N NaOH$)$. After removal of solvent under reduced pressure, the methohydroxide was transferred to a 10 mL microwave vessel, dried, and irradiated (microwave, $300 \mathrm{~W}, 140^{\circ} \mathrm{C}$ over 7.5 min , then an additional 4.5 min ). The crude product obtained was taken up in MeOH passed through a reversed-phase silica cartridge $\left(\mathrm{C}_{18}, 200 \mathrm{mg}\right)$ eluting with $\mathrm{MeOH}+0.1 \%$ TFA. Solvent was removed under a stream of $\mathrm{N}_{2}$, and the residue purified by HPLC (reversed-phase, $\mathrm{C}_{18}$ Phenomenex Luna, 5 $\mu \mathrm{m}, 10 \times 250 \mathrm{~mm}$, gradient: $40-100 \% \mathrm{CH}_{3} \mathrm{CN} / \mathrm{H}_{2} \mathrm{O}+0.1 \%$ TFA over 40 $\mathrm{min})$ to give the double-elimination product $24(\sim 140 \mu \mathrm{~g})$.

24: ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$, representative signals) $\delta 5.67$ (ddd, $J=17.2,10.2,7.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.62(\mathrm{dt}, J=17.2,10.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.35(\mathrm{~d}$, $J=10.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.29(\mathrm{~d}, J=17.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.94(\mathrm{brd}, J=17.2 \mathrm{~Hz}, 1 \mathrm{H})$, $2.93(\mathrm{~s}, 3 \mathrm{H}), 2.86(\mathrm{~s}, 3 \mathrm{H}), 2.58(\mathrm{t}, J=12.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.50(\mathrm{t}, J=12.2 \mathrm{~Hz}$, $1 \mathrm{H}), 0.98(\mathrm{~d}, J=7 \mathrm{~Hz}, 3 \mathrm{H})$; HRESIMS $m / z 487.4983[\mathrm{M}+\mathrm{H}]^{+}$(calcd for $\mathrm{C}_{33} \mathrm{H}_{63} \mathrm{~N}_{2}$, 487.4886).

Cross Metathesis of $\mathbf{2 4}$ with 2-Methoxy-6-vinylnaphthalene (13). Grubbs' second-generation catalyst ( $95 \mu \mathrm{~g}, 0.112 \mu \mathrm{~mol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.8 \mathrm{~mL})$ was added in three portions over a period of 11 h to alkene $24(140 \mu \mathrm{~g}, 0.199 \mu \mathrm{~mol})$ and 2-methoxy-6-vinylnaphthalene ${ }^{8}$ $(0.8 \mathrm{mg}, 4.20 \mu \mathrm{~mol})$, and the mixture was stirred vigorously in a sealed vial at $80^{\circ} \mathrm{C}$. The solvent was removed under a stream of $\mathrm{N}_{2}$, and the residue was passed through a reversed-phase silica cartridge $\left(\mathrm{C}_{18}, 200 \mathrm{mg}\right)$
eluting with $\mathrm{MeOH}+0.1 \%$ TFA, and finally purified by HPLC (reversedphase, $\mathrm{C}_{18}$ Phenomenex Luna, $5 \mu \mathrm{~m}, 10 \times 250 \mathrm{~mm}$, gradient: $40-100 \%$ $\mathrm{CH}_{3} \mathrm{CN} / \mathrm{H}_{2} \mathrm{O}+0.1 \%$ TFA over 40 min ) to give the 6-methoxy-2naphthylethenyl derivative $21(6 \mu \mathrm{~g})$ : $\mathrm{UV}\left(\mathrm{CH}_{3} \mathrm{CN}\right) \lambda 246,292 \mathrm{~nm}$; CD $\left(\mathrm{CH}_{3} \mathrm{CN}\right) \lambda 256(\Delta \varepsilon+3.9) ;{ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right.$, representative signals) $\delta 7.69(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.62(\mathrm{brs}, 1 \mathrm{H}), 7.55(\mathrm{dd}, J=8.6,2.0$ $\mathrm{Hz}, 1 \mathrm{H}), 7.19$ (brs, $J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.10(\mathrm{dd}, J=9.0,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.48(\mathrm{~d}$, $J=16.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.16(\mathrm{dd}, J=16.2,8.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.58(\mathrm{dt}, J=17.2,10.0 \mathrm{~Hz}$, $1 \mathrm{H}), 5.32(\mathrm{~d}, J=10.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.27(\mathrm{~d}, J=17.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.90(\mathrm{~s}, 3 \mathrm{H}), 2.89$ $(\mathrm{s}, 3 \mathrm{H}), 2.81(\mathrm{~s}, 3 \mathrm{H}), 1.12(\mathrm{~d}, J=7 \mathrm{~Hz}, 3 \mathrm{H})$; HRESIMS $m / z 643.5563$ $[\mathrm{M}]^{+}$, calcd 643.5561 for $\mathrm{C}_{44} \mathrm{H}_{71} \mathrm{~N}_{2} \mathrm{O}$.

Preparation of $p$-Bromophenacyl Derivatives. Bis-p-bromophenacyl Tetrahydroxestoproxamine A (25). The acetate salt of tetrahydroxestoproxamine A $(0.6 \mathrm{mg}, 1.06 \mu \mathrm{~mol})$ was converted to the free base 10 by addition of $\mathrm{KOH}(238 \mu \mathrm{~g}, 4.25 \mu \mathrm{~mol})$ in $\mathrm{MeOH}(0.5 \mathrm{~mL})$. After a few minutes the solvent was evaporated under a stream of $\mathrm{N}_{2}$ then on high vacuum. Toluene $(0.5 \mathrm{~mL})$ and $p$-bromophenacyl bromide $(5.9 \mathrm{mg}, 21.2$ $\mu \mathrm{mol})$ were added, and the mixture was stirred at $90^{\circ} \mathrm{C}$ for 12 h . The solvent was evaporated under a stream of $\mathrm{N}_{2}$, and the mixture was subjected to HPLC (reversed-phase Phenomenx, Luna, $\mathrm{C}_{18}, 10 \times 250 \mathrm{~mm} ; 2 \mathrm{~mL}$. $\min ^{-1} 40-100 \% \mathrm{CH}_{3} \mathrm{CN} / \mathrm{H}_{2} \mathrm{O}+0.1 \%$ TFA over 40 min ) to give the bis-p-bromophenacyl TFA salt $25(0.7 \mathrm{mg})$ : UV $(\mathrm{MeOH}) \lambda_{\max } 265(\log \varepsilon$ 4.38) nm; CD $(\mathrm{MeOH}) \lambda 254(\Delta \varepsilon+2.6), 276(\Delta \varepsilon-6.0) \mathrm{nm} ;{ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right.$, representative signals) $\delta 7.99(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.94$ $(\mathrm{d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.78(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.77(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 5.32$ $(\mathrm{d}, J=18.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.17(\mathrm{~d}, J=18.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.17(\mathrm{~d}, J=18.2 \mathrm{~Hz}, 1 \mathrm{H})$, $5.14(\mathrm{~s}, 2 \mathrm{H}), 4.18(\mathrm{brd}, J=11.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.11(\mathrm{brd}, J=12.7 \mathrm{~Hz}, 1 \mathrm{H}) ; 3.99$ $(\mathrm{m}, 1 \mathrm{H}), 3.98(\mathrm{~m}, 1 \mathrm{H}), 3.81(\mathrm{~m}, 2 \mathrm{H}), 3.77(\mathrm{~m}, 1 \mathrm{H}), 3.58(\mathrm{td}, J=13.1,3.3$ $\mathrm{Hz}, 1 \mathrm{H}), 3.56(\mathrm{~m}, 1 \mathrm{H}), 3.54(\mathrm{t}, J=12.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.31$ (under solvent), 3.24 $(\mathrm{t}, J=12.8 \mathrm{~Hz}, 1 \mathrm{H}), 1.21(\mathrm{q}, J=12.6 \mathrm{~Hz}, 1 \mathrm{H}) ;$ HRESIMS $m / z 838.3632$ $[\mathrm{M}]^{2+}$ (calcd for $\mathrm{C}_{46} \mathrm{H}_{68} \mathrm{Br}_{2} \mathrm{~N}_{2} \mathrm{O}_{2}, 838.3636$ ).

Bis-p-bromophenacyl Perhaliclonacyclamine (27). (-)-Perhaliclonacyclamine ${ }^{15}(\mathbf{2 6}, 150 \mu \mathrm{~g}, 0.317 \mu \mathrm{~mol})$ was stirred with $\mathrm{KOH}(71$ $\mu \mathrm{g}, 1.27 \mu \mathrm{~mol})$ in $\mathrm{MeOH}(0.5 \mathrm{~mL})$. After a few minutes the solvent was evaporated under a stream of $\mathrm{N}_{2}$, then on high vacuum. Toluene ( 0.5 mL ) and $p$-bromophenacyl bromide ( $1.8 \mathrm{mg}, 6.34 \mu \mathrm{~mol}$ ) were added, and the mixture was stirred at $90^{\circ} \mathrm{C}$ for 12 h . The solvent was evaporated under a stream of $\mathrm{N}_{2}$, and the mixture was subjected to HPLC (reversedphase Phenomenex, Luna, $\mathrm{C}_{18}, 10 \times 250 \mathrm{~mm} ; 2 \mathrm{~mL} \mathrm{~min}{ }^{-1}$; mobile phase/gradient: $40-100 \% \mathrm{CH}_{3} \mathrm{CN} / \mathrm{H}_{2} \mathrm{O}+0.1 \%$ TFA over 40 min ) to give the bis-p-bromophenacyl TFA salt $27(\mathrm{ca} .44 \mu \mathrm{~g}): \mathrm{UV}(\mathrm{MeOH}) \lambda$ $265 \mathrm{~nm} ; \mathrm{CD}(\mathrm{MeOH}) \lambda 254(\Delta \varepsilon+3.2), 272(-6.1) \mathrm{nm}$; HRESIMS $m /$ $z 866.3946[\mathrm{M}]^{2+}$ (calcd for $\left.\mathrm{C}_{48} \mathrm{H}_{72} \mathrm{Br}_{2} \mathrm{~N}_{2} \mathrm{O}_{2}, 866.3950\right)$.

Bis-p-bromophenacyl Perhydrohaliclonacyclamine E (29). A solution of $(+)$-haliclonacyclamine $E^{7}(9$, TFA salt, $0.8 \mathrm{mg}, 1.15 \mu \mathrm{~mol})$ in $\mathrm{MeOH} / \mathrm{AcOH}(49: 1,0.25 \mathrm{~mL})$ was stirred with $\mathrm{Pd} / \mathrm{C}(10 \%, 0.1 \mathrm{mg})$ at room temperature under $\mathrm{H}_{2}(1 \mathrm{~atm})$ for 48 h . The mixture was filtered through a nylon syringe filter $(0.45 \mu \mathrm{~m})$, and the solvent removed from the filtrate under a stream of $\mathrm{N}_{2}$. The residue was subjected to HPLC (Phenomenex, Synergi Hydro-RP, $4 \mu \mathrm{~m}, 10 \times 250 \mathrm{~mm} ; 2.5 \mathrm{~mL} \mathrm{~min}^{-1}$, 25:25:50:0.5 $\mathrm{CH}_{3} \mathrm{CN} / i-\mathrm{PrOH} / \mathrm{H}_{2} \mathrm{O} / \mathrm{TFA}$ ) to give perhydrohaliclonacyclamine E TFA salt ( $\mathbf{2 8}, 0.3 \mathrm{mg}$ ). The TFA salt was subjected to HPLC (Phenomenex Lux-cellulose, $1.5 \mathrm{~mL} \mathrm{~min}^{-1}$, 9:1:0.02 $\mathrm{CH}_{3} \mathrm{CN} / i-\mathrm{PrOH} /$ $\mathrm{Et}_{2} \mathrm{NH}$ ), and the solvent removed under a stream of $\mathrm{N}_{2}$ and further dried under high vacuum. Toluene ( 0.5 mL ) and $p$-bromophenacyl bromide $(1.2 \mathrm{mg}, 4.28 \mu \mathrm{~mol})$ were added, and the mixture was stirred at $90^{\circ} \mathrm{C}$ for 12 h . The solvent was removed under a stream of $\mathrm{N}_{2}$, and the mixture was subjected to HPLC (reversed-phase, Phenomenex, Luna, $\mathrm{C}_{18}, 10 \times 250$ $\mathrm{mm} ; 2 \mathrm{~mL} \mathrm{~min}^{-1}, 40-100 \% \mathrm{CH}_{3} \mathrm{CN} / \mathrm{H}_{2} \mathrm{O}+0.1 \%$ TFA over 40 min$)$ to give the bis-p-bromophenacyl TFA salt $29(\mathrm{ca} .7.0 \mu \mathrm{~g})$ : UV (MeOH) $\lambda 265$ $\mathrm{nm}(\log \varepsilon 4.38)$; CD $(\mathrm{MeOH}) \lambda 254(\Delta \varepsilon+2.6), 272(-6.0)$; HRESIMS $m / z 866.3940[M]^{2+}$ (calcd for $\mathrm{C}_{48} \mathrm{H}_{72} \mathrm{Br}_{2} \mathrm{~N}_{2} \mathrm{O}_{2}, 866.3950$ ).

Bis-p-bromophenacyl Dihydroxestoproxamine C(30). Xestoproxamine C free base (22, $0.4 \mathrm{mg}, 0.876 \mathrm{mmol})$ was stirred with $\mathrm{Pd} / \mathrm{C}(10 \%$,
$0.1 \mathrm{mg})$ in 0.25 mL of $\mathrm{MeOH} / \mathrm{AcOH}(49: 1)$ at room temperature under $\mathrm{H}_{2}(1 \mathrm{~atm})$ for 48 h . The mixture was filtered through a membrane filter $(0.45 \mathrm{~m})$, and the solvent was evaporated under a stream of $\mathrm{N}_{2}$ then by high vacuum. The dihydroxestoproxamine C acetate salt was subjected to HPLC (Phenomenex Lux-cellulose, $1.5 \mathrm{~mL} \min ^{-1}$, mobile phase: 9:1:0.02 $\mathrm{CH}_{3} \mathrm{CN} / i-\mathrm{PrOH} / \mathrm{Et}_{2} \mathrm{NH}$ ), and the solvent was concentrated under a stream of $\mathrm{N}_{2}$, then dried under reduced pressure. Dihydroxestoproxamine C free base was stirred with $p$-bromophenacyl bromide ( $3.0 \mathrm{mg}, 10.8 \mathrm{mmol}$ ) in toluene at $80^{\circ} \mathrm{C}$ for 2 h in a microwave reactor $(300 \mathrm{~W})$. The solvent was removed under a stream of $\mathrm{N}_{2}$, and the mixture subjected to RPHPLC (column: Phenomenx, Luna, C18(2), $10 \times 250 \mathrm{~mm} ; 2 \mathrm{~mL} \mathrm{~min}{ }^{-1}$; 40$100 \% \mathrm{CH}_{3} \mathrm{CN} / \mathrm{H}_{2} \mathrm{O}+0.1 \%$ TFA over 40 min ) to give the bis-pbromophenacyl TFA salt derivative $30(137 \mu \mathrm{~g})$ : UV $(\mathrm{MeOH}) \lambda_{\text {max }} 265$ $\mathrm{nm} ; \mathrm{CD}(\mathrm{MeOH}) \lambda 254(\Delta \varepsilon+3.5), 276(\Delta \varepsilon-7.1) \mathrm{nm} ;$ HRESIMS $m / z$ $852.3792[\mathrm{M}]^{2+}$ (calcd for $\mathrm{C}_{47} \mathrm{H}_{70} \mathrm{Br}_{2} \mathrm{~N}_{2} \mathrm{O}_{2}, 852.3799$ ).

Synthesis of Model Compound 20. Amide 15. $n$ - BuLi (2.21 M in hexanes, $3.0 \mathrm{~mL}, 6.64 \mathrm{mmol}$ ) was added to a stirred solution of anhydrous lithium chloride $(0.84 \mathrm{~g}, 19.8 \mathrm{mmol})$ and diisopropylamine $(0.94 \mathrm{~mL}, 6.64 \mathrm{mmol})$ in THF $(20 \mathrm{~mL})$ at $-78^{\circ} \mathrm{C}$. The mixture was kept at $-78^{\circ} \mathrm{C}$ for 20 min , then placed in an ice bath for 30 min , then cooled back to $-78{ }^{\circ} \mathrm{C}$. Amide 16 ( $0.70 \mathrm{~g}, 3.16 \mathrm{mmol}$, prepared by acylation of pseudoephedrine with propionyl chloride) was added over 10 min , and the solution was slowly warmed to room temperature over 1.5 h and stirred an additional 15 min , then cooled to $-40^{\circ} \mathrm{C}$. The $1-\mathrm{O}-$ TBS-9-iododecane 17 ( $1.75 \mathrm{~g}, 4.39 \mathrm{mmol}$ ) was added dropwise, and the mixture was allowed to warm to room temperature overnight with stirring. The mixture was poured into saturated $\mathrm{NH}_{4} \mathrm{Cl}$ solution ( 75 mL ), and the organic layer was separated. The aqueous layer was extracted with $\mathrm{EtOAc}(4 \times 50 \mathrm{~mL})$, the combined organic layers were dried $\left(\mathrm{MgSO}_{4}\right)$, the solvent was removed under reduced pressure, and the crude product was purified by flash chromatography $\left(\mathrm{SiO}_{2}, 1: 3\right.$ $\mathrm{EtOAc} /$ hexanes $)$ to give the amide $15\left(1.20 \mathrm{~g}, 77 \%\right.$ yield): $[\alpha]_{\mathrm{D}}^{22}-36.7$ (c 3.1, $\mathrm{CHCl}_{3}$ ); FTIR (ATR) v 3376, 2926, 2854, 1620, 1464, 1408, 1254, 1099, 1052, 835, 775, $701 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $7.38-7.23(\mathrm{~m}, 5 \mathrm{H}), 4.61(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}$, major $), 4.59(\mathrm{~m}, 1 \mathrm{H}$, minor $)$, 4.39 (brs, 1 H, major), $4.08(\mathrm{p}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}$, minor), $3.59(\mathrm{t}, J=6.4 \mathrm{~Hz}$, 2 H, major), $3.58(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}$, minor), $2.91(\mathrm{~s}, 3 \mathrm{H}$, minor $), 2.84(\mathrm{~s}$, 3 H, major), $2.78(\mathrm{t}, J=6.4 \mathrm{~Hz}, 1 \mathrm{H}$, minor), 2.58 ( sex, $J=7.2 \mathrm{~Hz}, 1 \mathrm{H}$, major), 2.29 (brs, 1 H, minor), 1.71 (brs, 1 H , major), $1.60-1.46$ (m), $1.31-1.19(\mathrm{~m}), 1.14(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}$, major), $1.08(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}$, major), $1.02\left(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}\right.$, minor), $0.89(\mathrm{~s}, 9 \mathrm{H}), 0.04(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 179.3$ (C), 142.6 (C), 128.3 (CH), 127.5 $(\mathrm{CH}), 126.3(\mathrm{CH}), 76.5(\mathrm{CH}), 63.3\left(\mathrm{CH}_{2}\right), 36.6(\mathrm{CH}), 34.0\left(\mathrm{CH}_{2}\right)$, $32.9\left(\mathrm{CH}_{2}\right), 29.9-29.4\left(5 \times \mathrm{CH}_{2}\right), 27.4\left(\mathrm{CH}_{2}\right), 26.0\left(3 \times \mathrm{CH}_{3}\right), 25.8$ $\left(\mathrm{CH}_{2}\right), 18.4(\mathrm{C}), 17.3\left(\mathrm{CH}_{3}\right), 14.5\left(\mathrm{CH}_{3}\right),-5.3\left(2 \times \mathrm{CH}_{3}\right)$; HRESIMS $m / z 514.3685[\mathrm{M}+\mathrm{Na}]^{+}$(calcd for $\left.\mathrm{C}_{29} \mathrm{H}_{53} \mathrm{NO}_{3} \mathrm{SiNa}, 514.3687\right)$.

Alcohol 18. $\mathrm{BH}_{3} \cdot$ THF complex ( 1 M in THF, $3.0 \mathrm{~mL}, 3 \mathrm{mmol}$ ) was added to pyrrolidine $(246 \mathrm{~mL}, 3 \mathrm{mmol})$ at $0^{\circ} \mathrm{C}$, and the solution was warmed to room temperature and stirred for 45 min . The solution was cooled to $0^{\circ} \mathrm{C}$ and treated, dropwise, with $n-\operatorname{BuLi}(2.21 \mathrm{M}$ in hexanes, $1.36 \mathrm{~mL}, 3 \mathrm{mmol}$ ) with stirring for an additional 30 min . The amide 15 ( $500 \mathrm{mg}, 1 \mathrm{mmol}$ ) in THF $(10 \mathrm{~mL})$ was added and stirred at room temperature overnight. $\mathrm{HCl}(3 \mathrm{M}, 5 \mathrm{~mL})$ was added to quench the excess hydride, and the layers were separated. Ether ( 5 mL ) was added to the aqueous layer, and the mixture cooled to $0^{\circ} \mathrm{C}$ before being made basic ( $\mathrm{pH} 9-10$ ) by addition of 2 N NaOH . The aqueous mixture was extracted with ether $(3 \times 5 \mathrm{~mL})$, and the combined organic extracts were washed with $1: 1$ brine $/ 1 \mathrm{~N} \mathrm{NaOH}(2 \times 10 \mathrm{~mL})$ and dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$. The solvent was evaporated and subjected to flash chromatography $\left(\mathrm{SiO}_{2}, 12.5 \% \mathrm{EtOAc} /\right.$ hexanes $)$ to give the alcohol $18(135 \mathrm{mg}, 41 \%$ yield): $[\alpha]_{\mathrm{D}}^{23}-5.5\left(c 2.1, \mathrm{CHCl}_{3}\right)$; FTIR (ATR) $v 3340,2926,2855$, 1464, 1254, 1102, 1041, 835, $775 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H} \mathrm{NMR} \mathrm{( } 400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ $3.59(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.50(\mathrm{dd}, J=10.0,5.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.41(\mathrm{dd}, J=10.0$, $6.8 \mathrm{~Hz}, 1 \mathrm{H}), 1.59(\mathrm{~m}, 1 \mathrm{H}), 1.50(\mathrm{p}, J=6.8 \mathrm{~Hz}, 2 \mathrm{H}), 1.41-1.26$
$(\mathrm{m}, 15 \mathrm{H}), 1.09(\mathrm{~m}, 1 \mathrm{H}) 0.91(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}), 0.89(\mathrm{~s}, 9 \mathrm{H}), 0.04(\mathrm{~s}, 6 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 68.4\left(\mathrm{CH}_{2}\right), 63.3\left(\mathrm{CH}_{2}\right), 35.7(\mathrm{CH}), 33.2$ $\left(\mathrm{CH}_{2}\right), 32.9\left(\mathrm{CH}_{2}\right), 29.9-29.4\left(5 \times \mathrm{CH}_{2}\right), 27.0\left(\mathrm{CH}_{2}\right), 26.0\left(3 \times \mathrm{CH}_{3}\right)$, $25.8\left(\mathrm{CH}_{2}\right), 18.4(\mathrm{C}), 16.6\left(\mathrm{CH}_{3}\right),-5.3\left(2 \times \mathrm{CH}_{3}\right)$; HRESIMS $\mathrm{m} / \mathrm{z}$ $353.2849[\mathrm{M}+\mathrm{Na}]^{+}$(calcd for $\left.\mathrm{C}_{19} \mathrm{H}_{42} \mathrm{O}_{2} \mathrm{SiNa}, 353.2846\right)$.

Aldehyde 19a. DMSO ( $64 \mathrm{~mL}, 0.91 \mathrm{mmol}$ ) was added dropwise to a stirred solution of oxalyl chloride ( $51 \mathrm{~mL}, 0.60 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $(3 \mathrm{~mL})$ at $-78{ }^{\circ} \mathrm{C}$ and stirred for 10 min . A solution of the alcohol 18 ( $100 \mathrm{mg}, 0.302 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ was added, and the mixture was stirred for 30 min at $-50^{\circ} \mathrm{C} . \mathrm{Et}_{3} \mathrm{~N}(168 \mathrm{~mL}, 1.21 \mathrm{mmol})$ was added, and the mixture was brought to $-30^{\circ} \mathrm{C}$ followed by stirring an additional 30 $\min$. Saturated $\mathrm{NH}_{4} \mathrm{Cl}$ solution $(5 \mathrm{~mL})$ was added, and the layers were separated. The aqueous layer was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \times 3 \mathrm{~mL})$, the combined organic extracts were dried $\left(\mathrm{MgSO}_{4}\right)$ and concentrated, and the residue was purified by flash chromatography $\left(\mathrm{SiO}_{2}, 3 \% \mathrm{EtOAc} /\right.$ hexanes) to give the aldehyde $19 \mathrm{a}\left(91 \mathrm{mg}, 92 \%\right.$ yield): $[\alpha]_{\mathrm{D}}^{24}+12.8$ (c 3.4, $\mathrm{CHCl}_{3}$ ); FTIR (ATR) $v 2926,2854,1730,1463,1254,1099,835,774$ $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.61(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.59(\mathrm{t}, \mathrm{J}=$ $6.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.32(\mathrm{sd}, J=7.2,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 1.69(\mathrm{~m}, 1 \mathrm{H}), 1.50(\mathrm{p}, J=7.2 \mathrm{~Hz}$, $2 \mathrm{H}), 1.25(\mathrm{~m}, 15 \mathrm{H}), 1.08(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}), 0.89(\mathrm{~s}, 9 \mathrm{H}), 0.04(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 205.5(\mathrm{CH}), 63.3\left(\mathrm{CH}_{2}\right), 46.3(\mathrm{CH}), 32.9$ $\left(\mathrm{CH}_{2}\right), 30.5\left(\mathrm{CH}_{2}\right), 29.6-29.4\left(5 \times \mathrm{CH}_{2}\right), 26.9\left(\mathrm{CH}_{2}\right), 26.0\left(3 \times \mathrm{CH}_{3}\right)$, $25.8\left(\mathrm{CH}_{2}\right), 18.4(\mathrm{C}), 13.3\left(\mathrm{CH}_{3}\right),-5.3\left(2 \times \mathrm{CH}_{3}\right)$; HRESIMS $\mathrm{m} / \mathrm{z}$ $351.2691[\mathrm{M}+\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{19} \mathrm{H}_{40} \mathrm{O}_{2} \mathrm{SiNa}, 351.2690$ ).

Olefin 19. $n-\mathrm{BuLi}(1.2 \mathrm{M}$ in hexanes, $143 \mathrm{uL}, 0.171 \mathrm{mmol})$ was added to a slurry of $\mathrm{MePPh}_{3} \mathrm{Br}(65 \mathrm{mg}, 0.183 \mathrm{mmol})$ in THF $(0.5 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$; then the mixture was stirred at room temperature for 5 min . The aldehyde 19a ( $40 \mathrm{mg}, 0.122 \mathrm{mmol}$ ) in THF $(0.5 \mathrm{~mL})$ was added, and stirring continued for 2 h at room temperature. Saturated $\mathrm{NH}_{4} \mathrm{Cl}$ solution $(1 \mathrm{~mL})$ was added followed by ether $(2 \mathrm{~mL})$, and the layers were separated. The aqueous layer was extracted with ether $(2 \times 2 \mathrm{~mL})$, the combined organic extracts were dried $\left(\mathrm{MgSO}_{4}\right)$ and concentrated, and the residue was purified by flash chromatography $\left(\mathrm{SiO}_{2}\right.$, hexanes $)$ to give the chiral olefin 19 ( $35 \mathrm{mg}, 88 \%$ yield): $[\alpha]^{23}{ }_{\mathrm{D}}+6.5\left(c 2.8, \mathrm{CHCl}_{3}\right)$; FTIR (ATR) $v 2925,2854,1463,1254,1099,994,909,834,773 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR (400 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 5.69$ (ddd, $\left.J=17.2,10.0,7.2 \mathrm{~Hz}, 1 \mathrm{H}\right), 4.94$ (ddd, $J=17.2,2.0,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.89(\mathrm{ddd}, J=10.0,2.4,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.51(\mathrm{t}$, $J=6.8 \mathrm{~Hz}, 2 \mathrm{H}), 2.10(\mathrm{~m}, 1 \mathrm{H}), 1.50(\mathrm{p}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.25(\mathrm{~m}, 16 \mathrm{H}), 0.97$ $(\mathrm{d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}), 0.89(\mathrm{~s}, 9 \mathrm{H}), 0.05(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $(100 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta 145.1(\mathrm{CH}), 112.2\left(\mathrm{CH}_{2}\right), 63.3\left(\mathrm{CH}_{2}\right), 37.8(\mathrm{CH}), 36.7\left(\mathrm{CH}_{2}\right)$, $32.9\left(\mathrm{CH}_{2}\right), 29.8-29.4\left(5 \times \mathrm{CH}_{2}\right), 27.2\left(\mathrm{CH}_{2}\right), 26.0\left(3 \times \mathrm{CH}_{3}\right), 25.8$ $\left(\mathrm{CH}_{2}\right), 20.2\left(\mathrm{CH}_{3}\right), 18.4(\mathrm{C}),-5.3\left(2 \times \mathrm{CH}_{3}\right)$; HRESIMS $m / z 327.3081$ $[\mathrm{M}+\mathrm{H}]^{+}$(calcd for $\mathrm{C}_{20} \mathrm{H}_{43} \mathrm{O}_{2} \mathrm{SiNa}$, 327.3078).

Alkene 20. A mixture of alkene $19(2 \mathrm{mg}, 6.12 \mathrm{mmol})$ and 2-methoxy-6-vinylnaphthalene $(11.3 \mathrm{mg}, 61.2 \mathrm{mmol})$ was treated with Grubbs' second-generation catalyst ( $1 \mathrm{mg}, 1.22 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ (3 $\mathrm{mL})$ at $70^{\circ} \mathrm{C}$ in a sealed vial for 12 h . Additional catalyst ( $1 \mathrm{mg}, 1.22$ mmol ) was added, and the mixture was stirred an additional 6 h at $80^{\circ} \mathrm{C}$. The solvent was removed under a stream of $\mathrm{N}_{2}$, and the mixture was passed through a reversed-phase $\left(\mathrm{C}_{18}\right)$ silica cartridge $\left(\mathrm{CH}_{3} \mathrm{CN}\right.$ then MeOH ), followed by HPLC (reversed-phase, $\mathrm{C}_{8}, 85 \% \mathrm{CH}_{3} \mathrm{CN} / \mathrm{H}_{2} \mathrm{O}$ ) to give alkene $20(1.3 \mathrm{mg}, 42 \%):[\alpha]_{\mathrm{D}}^{22}+31.1\left(c 1.0, \mathrm{CHCl}_{3}\right)$; UV $\left(\mathrm{CH}_{3} \mathrm{CN}\right) 246 \mathrm{~nm}(\log \varepsilon 4.62), 292$ (4.20); CD $\left(\mathrm{CH}_{3} \mathrm{CN}\right) \lambda 255 \mathrm{~nm}$ $(\Delta \varepsilon+3.8)$; FTIR (ATR) $v 2925,2853,1256,1100,1035,837,777 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.68(\mathrm{~d}, J=9.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.66(\mathrm{~d}, J=9.0$, $1 \mathrm{H}), 7.62(\mathrm{brs}, 1 \mathrm{H}), 7.55(\mathrm{dd}, J=9.0,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.11(\mathrm{dd}, J=9.0 .2 .0$ $\mathrm{Hz}, 1 \mathrm{H}), 7.10$ (brs, 1H), $6.46(\mathrm{~d}, J=16 \mathrm{~Hz}, 1 \mathrm{H}), 6.16$ (dd, $J=16.0,8.5$ $\mathrm{Hz}, 1 \mathrm{H}), 3.91(\mathrm{~s}, 3 \mathrm{H}), 3.59(\mathrm{t}, J=7.0 \mathrm{~Hz}, 2 \mathrm{H}), 2.32(\mathrm{sep}, J=6.5 \mathrm{~Hz}, 1 \mathrm{H})$, $1.50(\mathrm{~m}, 2 \mathrm{H}), 1.38(\mathrm{~m}, 2 \mathrm{H}), 1.27(\mathrm{~m}, 14 \mathrm{H}), 1.10(\mathrm{~d}, J=6.5 \mathrm{~Hz}, 3 \mathrm{H}), 0.88$ ( $\mathrm{s}, 9 \mathrm{H}$ ), $0.04(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 157.4$ (C), 136.6 ( CH ), 133.7 (C), 133.4 (C), 129.3 (CH), 129.1 (C), 128.0 (CH), 126.9 $(\mathrm{CH}), 125.1(\mathrm{CH}), 124.2(\mathrm{CH}), 118.8(\mathrm{CH}), 105.8(\mathrm{CH}), 63.3\left(\mathrm{CH}_{2}\right)$, $55.3\left(\mathrm{CH}_{3}\right), 37.4(\mathrm{CH}), 37.2\left(\mathrm{CH}_{2}\right), 32.9\left(\mathrm{CH}_{2}\right), 29.8-29.4\left(5 \times \mathrm{CH}_{2}\right)$, $27.4\left(\mathrm{CH}_{2}\right), 26.0\left(3 \times \mathrm{CH}_{3}\right), 25.8\left(\mathrm{CH}_{2}\right), 20.8\left(\mathrm{CH}_{3}\right), 18.4(\mathrm{C}),-5.3$
$\left(2 \times \mathrm{CH}_{3}\right) ;$ HRESIMS $m / z 482.3573[\mathrm{M}]^{+}$(calcd for $\mathrm{C}_{31} \mathrm{H}_{50} \mathrm{O}_{2} \mathrm{Si}$, 482.3575).

## ■ ASSOCIATED CONTENT

s Supporting Information. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR and 2 D NMR spectra of $1-3,{ }^{13} \mathrm{C}$ NMR spectra of $\mathbf{1 0}, 11,22$, and $23,{ }^{1} \mathrm{H}$ NMR spectra of $12,14,24,25$, and 30 , and ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR of all new synthetic intermediates $(\mathbf{1 5}, \mathbf{1 8}, \mathbf{1 9 a}, 19$, and 20$)$. This material is available free of charge via the Internet at http://pubs.acs.org.

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## - DEDICATION

Dedicated to Dr. Koji Nakanishi of Columbia University for his pioneering work on bioactive natural products.

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(15) The absolute configuration of ( - -perhaliclonacyclamine was established from X-ray single-crystal structure determination of the parent compound, ( + )-tetradehydrohaliclonacyclamine A, obtained with $\mathrm{Cu} \mathrm{K} \alpha$ radiation and interpretation of anomalous scattering (Bijvoet method). Mudianta, I. W.; Katavic, P. L.; Lambert, L. K.; Hayes, P. Y.; Banwell, M. G.; Munro, M. H. G.; Bernhardt, P. V.; Garson, M. J. Tetrahedron 2010, 66, 2752-2760.
(16) It is worth recalling that it is not possible, strictly, to assign absolute configuration to chiral compounds by comparison of signs of $[\alpha]_{\mathrm{D}}$, alone, except for enantiomers, even those as similar as $1-3$ and 26. Two counter-examples are prescient and sufficient to illustrate this often ignored tenet in stereochemistry. The paired compounds ( - )perhydrohaliclonacyclamine (26) and the parent $(+)$-tetradehydrohaliclonacyclamine A (31), and ( - )-haliclonacyclamine A $(6)\left([\alpha]_{\mathrm{D}}=-\right.$ $3.4)$ and $(+)$-haliclonacyclamine $B(7)\left([\alpha]_{D}=+3.4\right)$, differ only in the position of a single double bond in the lower linking chain, yet the pairs show opposite signs of specific rotation. Also, the absolute configuration of $\left.(-)-26\left([\alpha]_{\mathrm{D}}=-20.9\right]\right)^{15}$ was shown to be antipodal to that of $(-)$ 6 despite the same sign of $[\alpha]_{\mathrm{D}}$. Lastly, the signs of $[\alpha]_{\mathrm{D}}$ of alkaloid salts and their parent free bases may also differ.
(17) We are grateful to Professor Mary Garson, University of Queensland, for the generous gift of 26.
(18) We are grateful to Professor Roberto Berlinck, University of São Paulo, for the generous gift of 9 .
(19) The chair-chair conformations and N-configurations adopted by the bis-piperidine ring systems in the pairs 25 and 29 and 27 and 30 are preserved with respect to their starting materials (see Supporting Information for assignment of conformation of 25 using NOESY data). Although the Br-phenacyl substitutents are disposed differently-equatorial/axial on rings A and B of 25 and 29 and axial/axial for 27 and 30 -examination of molecular models shows the $\mathrm{C}-\mathrm{N}$ bond vectors subtend similar angles $(\theta$ $\sim 90^{\circ}$ ) in all four derivatives. This explains why the diastereomeric variants give the same CD spectra, because helicity and the consequent exciton coupling is dictated largely by the orthogonality of the piperidine ring planes. It is likely that the reactive conformers of each bis-piperidine ring require an equatorial lone pair and, consequently, a ring flip for the $B$ ring in all four starting materials and an A ring flip in 22 then 26 (see Figure 2). Reaction of 22, with both lone pairs on N in the axial orientation, was particularly sluggish and required more forcing conditions (microwave reactor, $140^{\circ} \mathrm{C}$ ) compared to 10,26 , and $28\left(90^{\circ} \mathrm{C}\right)$, possibly a consequence of torsional constraints imposed on 22 for the A ring flip (two less $\mathrm{CH}_{2}$ groups in the "northern" linker chain) compared to the more flexible 26.
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